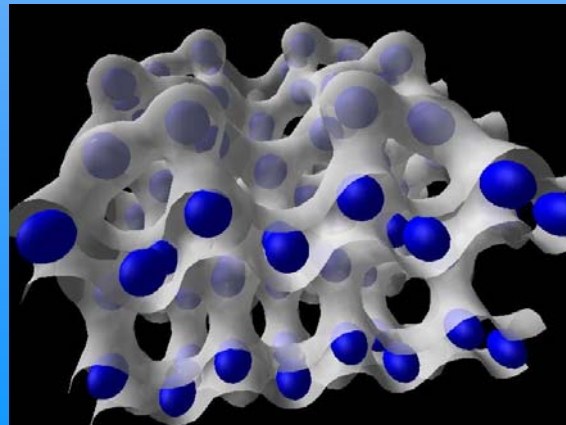
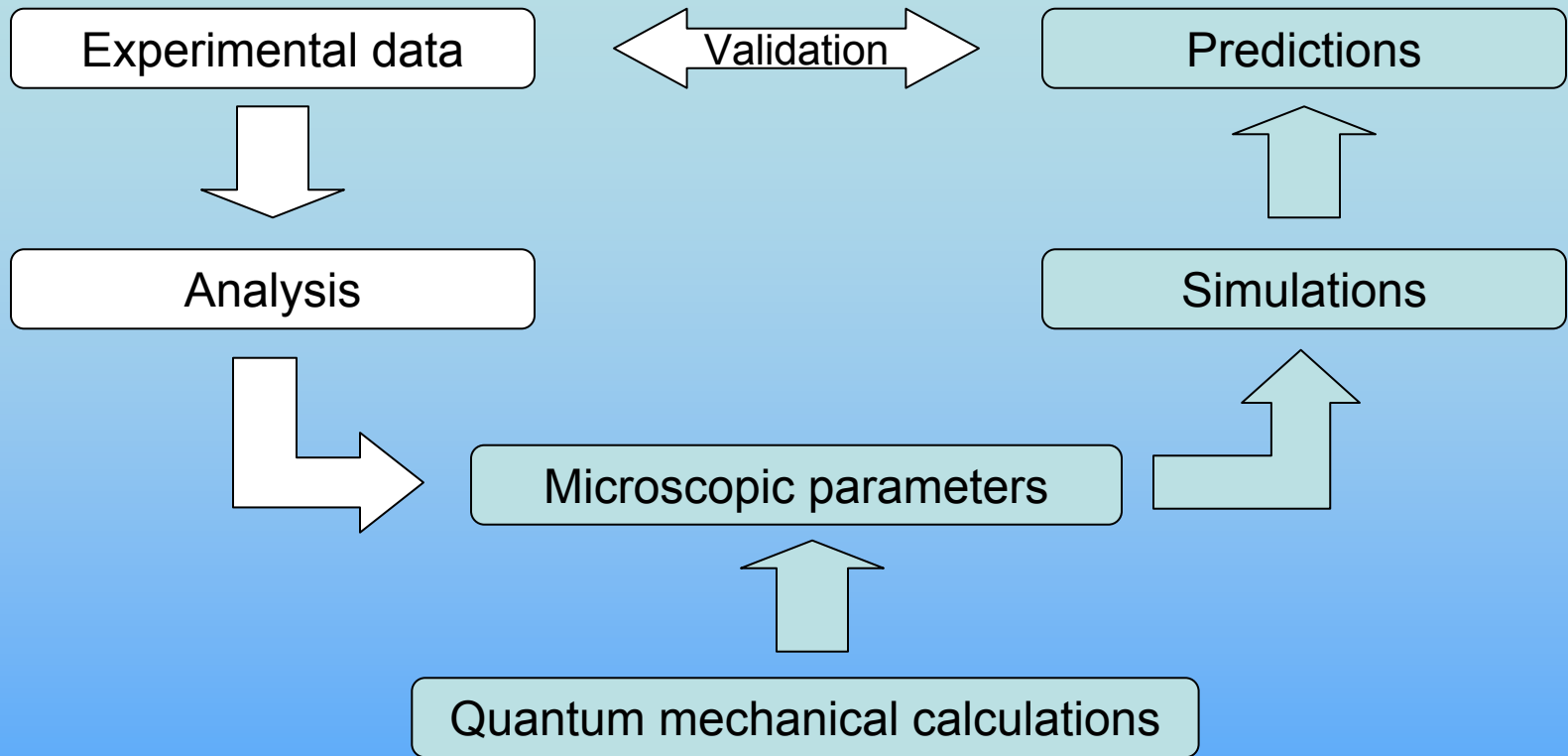
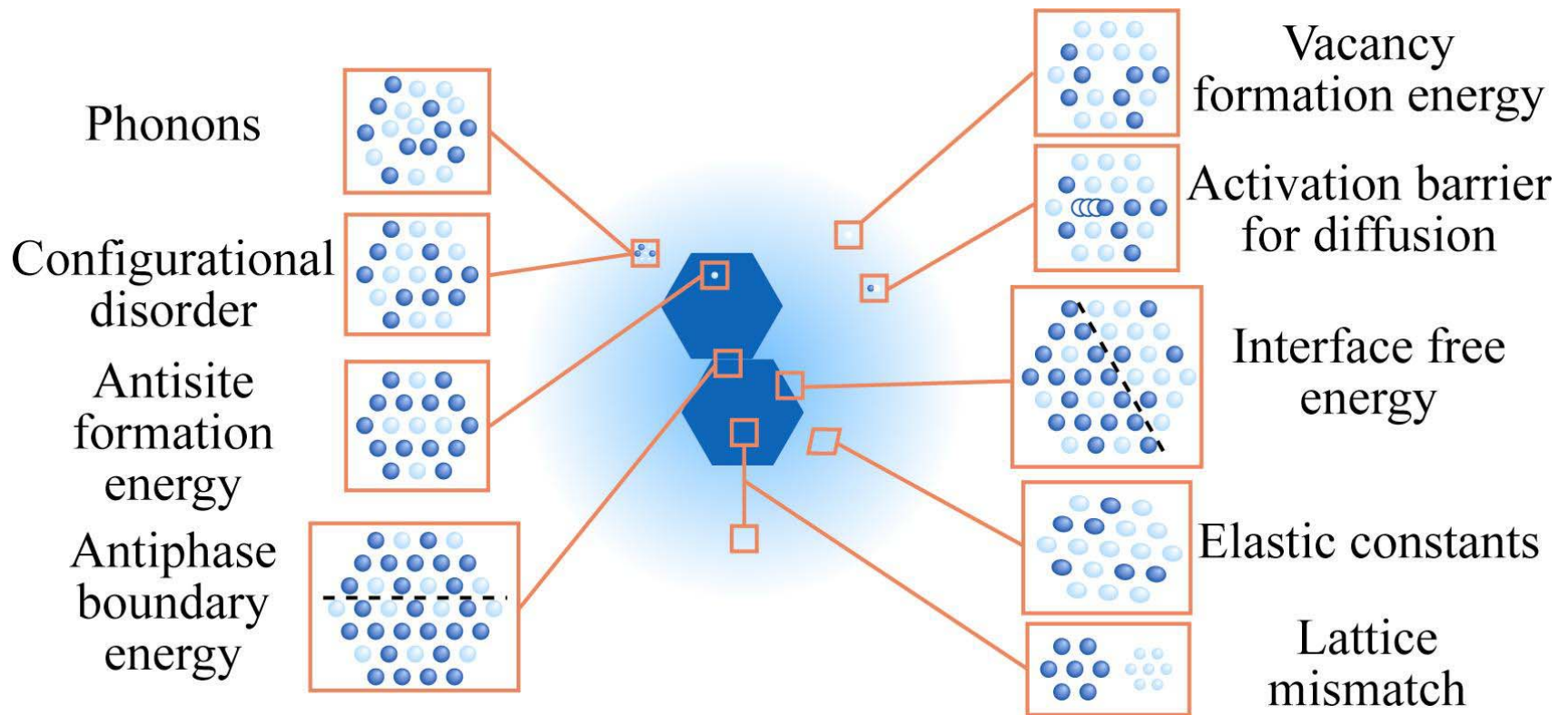


# Traditional vs. *ab initio* modeling



First-principles (or *ab initio*) calculations based on Density Functional Theory (DFT).

# Microscopic parameters governing materials properties

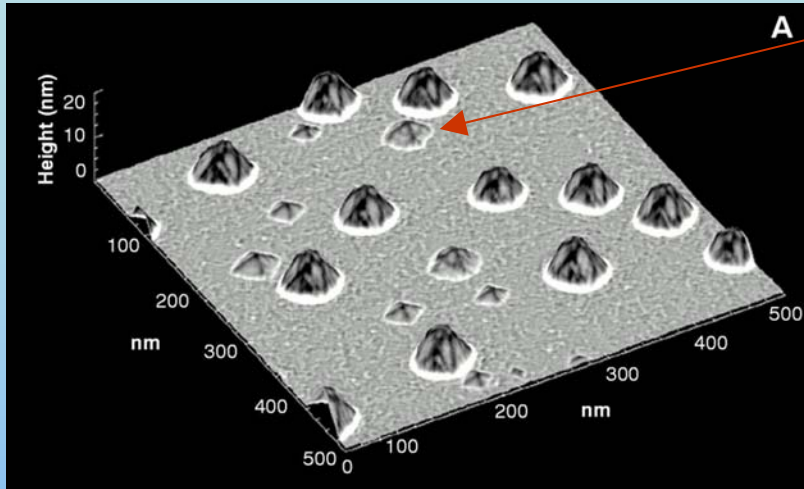


Example: nucleation and growth of precipitate

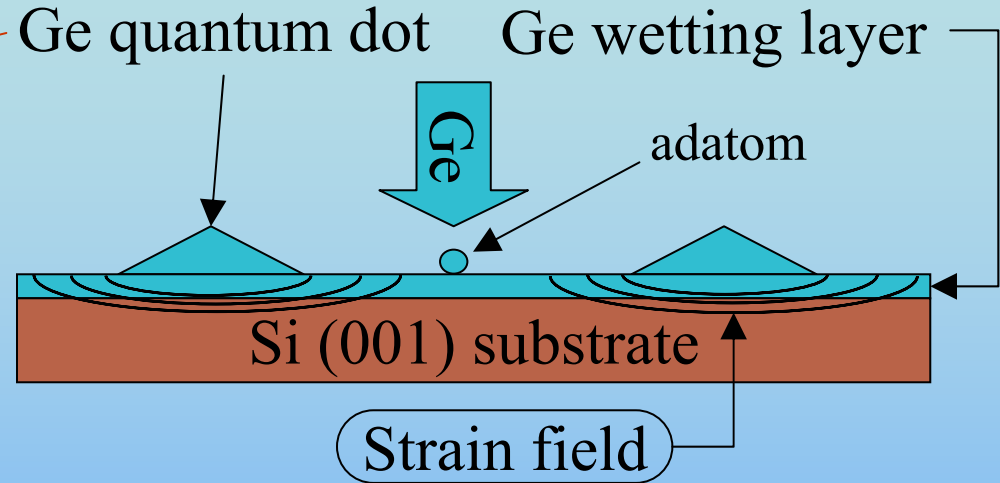
# Outline

- Example of *ab initio* calculations
- Why *ab initio* Thermodynamics?
- Methods
- Applications

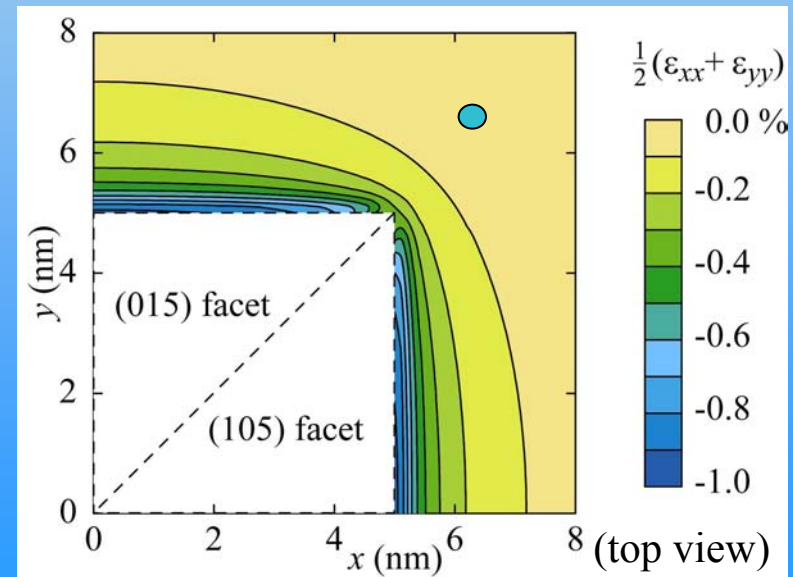
# Self-Assembling Ge/Si(001) Quantum Dots



(Medeiros-Ribeiro, *et al.*, 1998)



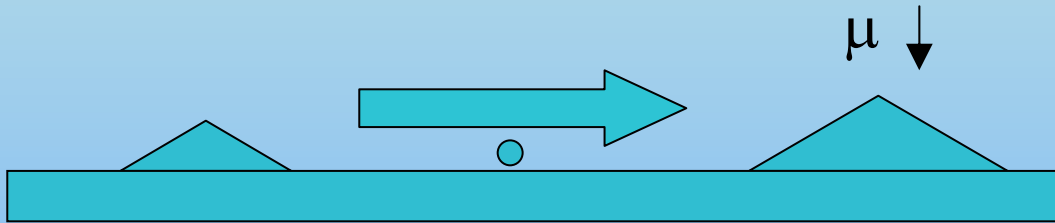
Goal: quantify the **strain-dependence** of the parameters governing **Ge adatom diffusion**



**Related works:** Roland and Gilmer (1992), Spjut and Faux (1994), Schroeder and Wolf (1997), Ratsch *et al.* (1997), Zoethout *et al.* (2000), Shu *et al.* (2001), Penev *et al.* (2001).

# Unwanted Coarsening

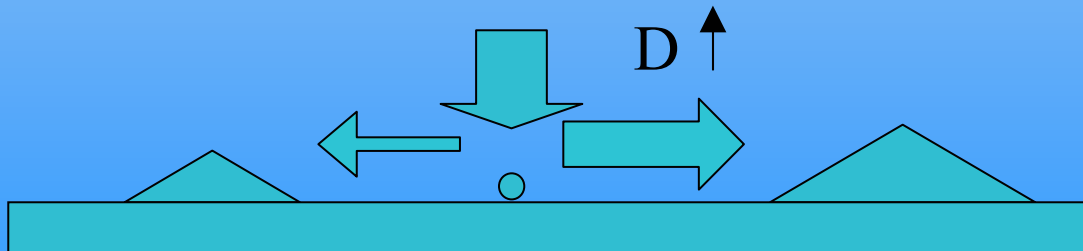
- Thermodynamic effect



**Larger island:**

Lower surface/volume  
(capillarity)

- Kinetic effect (?)



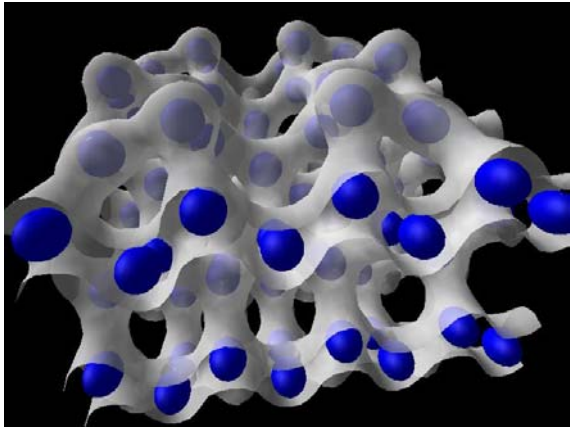
**Near Larger island:**

Higher adatom mobility

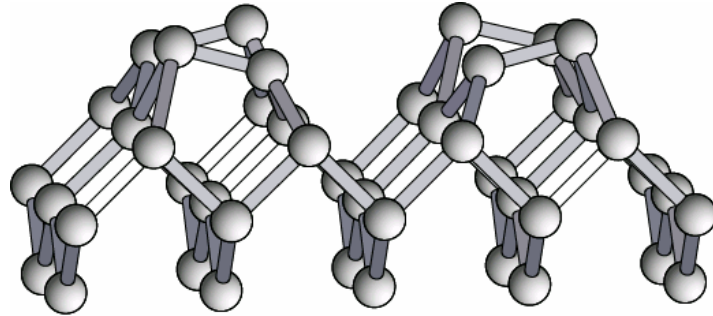
(Lower barrier for  
adatom attachment)

e.g. Koduvally and Zangwill (1999)

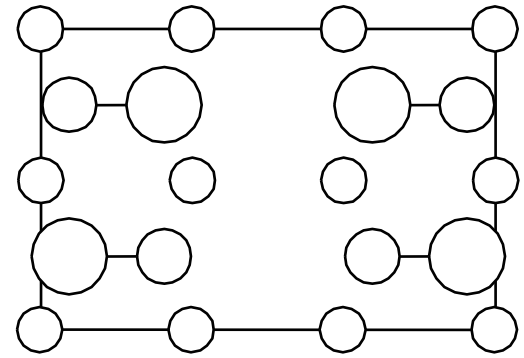
# Ge (001) Surface c(4x2) Reconstruction



isoelectronic density plot



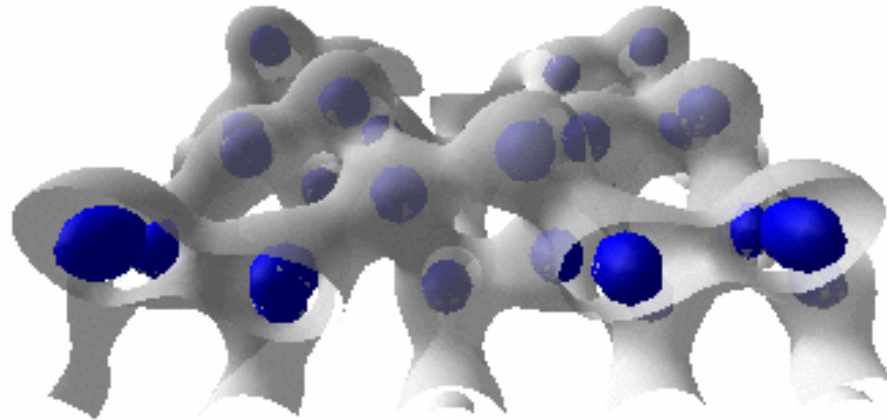
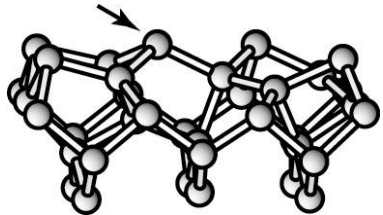
3D schematic



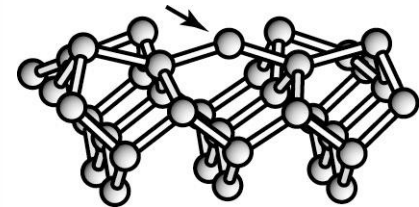
Schematic top view

## Ge Adatom Diffusion on Ge (001)

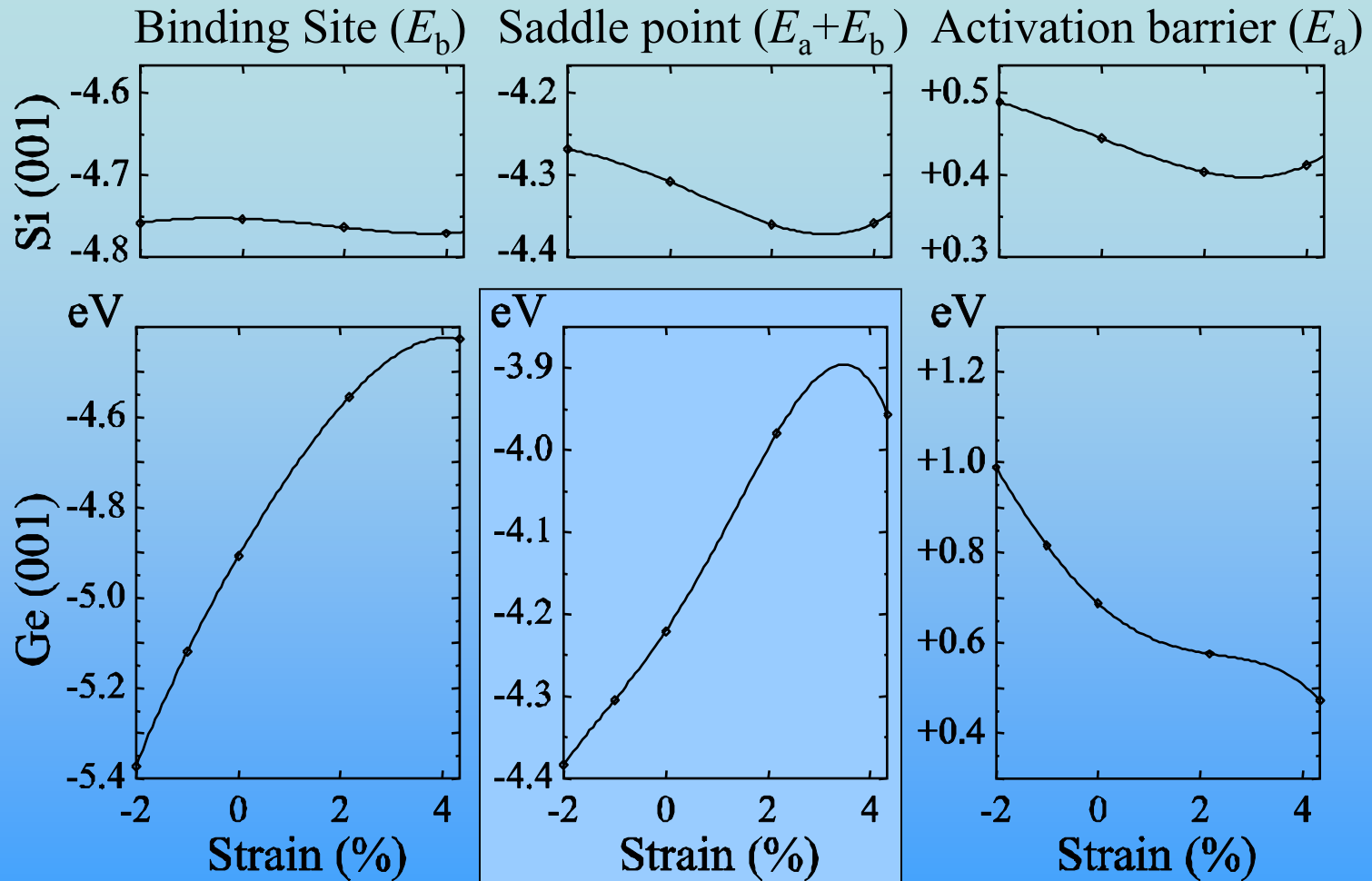
Binding site



Saddle point



# Strain dependence



Diffusion-driven kinetic coarsening

0 % strain = Si lattice param.      4 % strain = Ge lattice param.

# Magnitude of the effect

Reference	System adatom/subst.	$dE_b/de$ (eV/%)	$dE_a/de$ (eV/%)
Roland <i>et al.</i> (1992)	Si/Si(001)		-0.023
Ratsch <i>et al.</i> (1997)	Ag/Ag(111)	-0.028	0.007
Zoethout <i>et al.</i> (2000)	Si <sub>2</sub> /Si(001)		0.038
Penev <i>et al.</i> (2001)	In/GaAs(001)	-0.050	-0.020
Shu <i>et al.</i> (2001)	Si/Si(001)		-0.050
Present	Ge/Si(001)	0.002	-0.022
Present	Ge/Ge(001)/Si	0.212	-0.128



# The “Virtual” Laboratory

## Advantages

- Absolute control over “experimental” conditions
- Unlimited “characterization” capabilities
- Ability to perform thought experiments
- Automation of repetitive tasks
- “Materials by design”

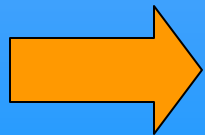
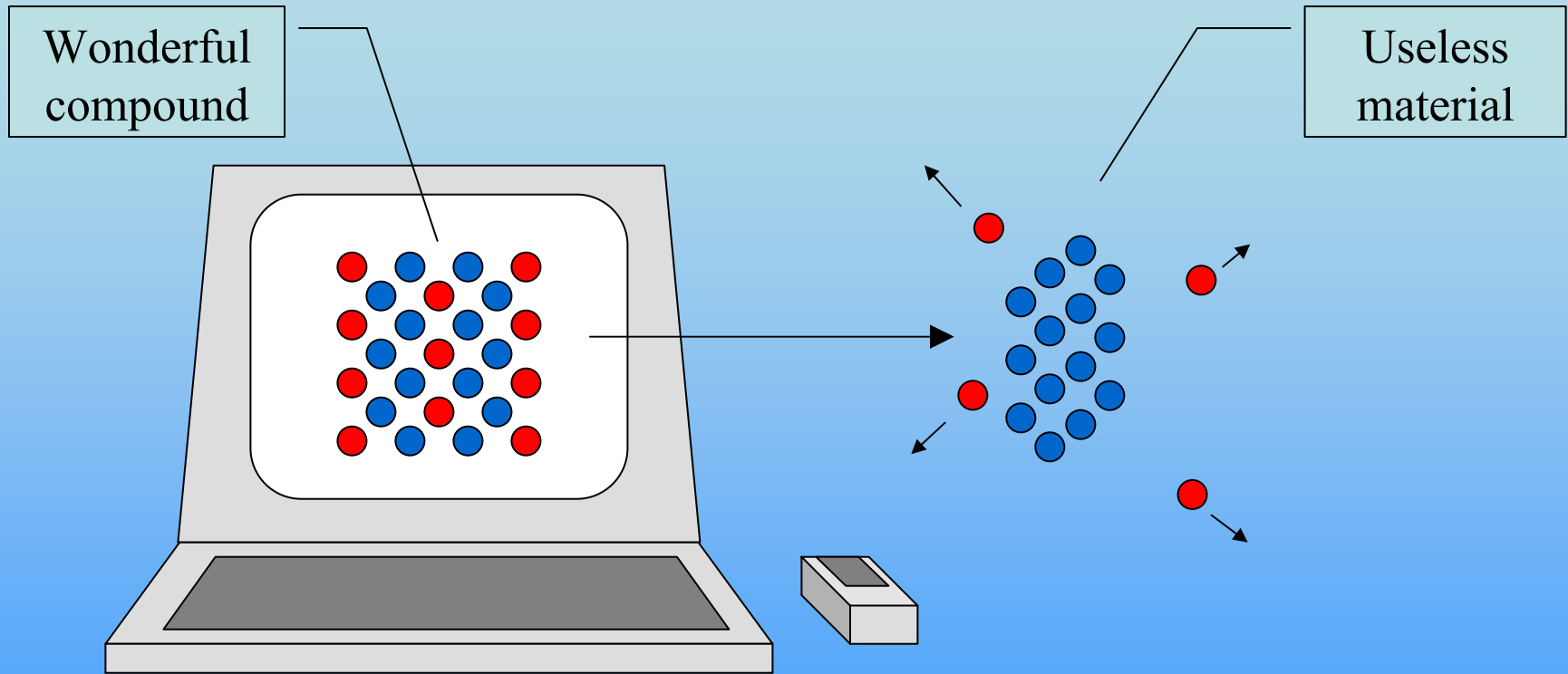
## Weaknesses

- Limited by computational resources
- Cannot predict effects not included in simulation

# Outline

- Example of *ab initio* calculations
- Why *ab initio* Thermodynamics?
- Methods
- Applications

# From a virtual to a real material...

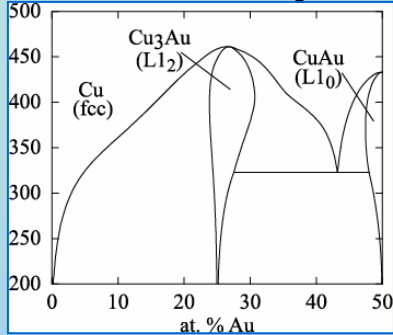


We need a way to predict phase stability:  
First-principles phase diagram calculations

# Outline

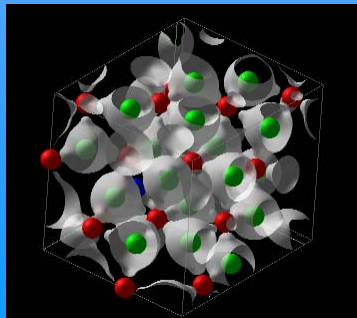
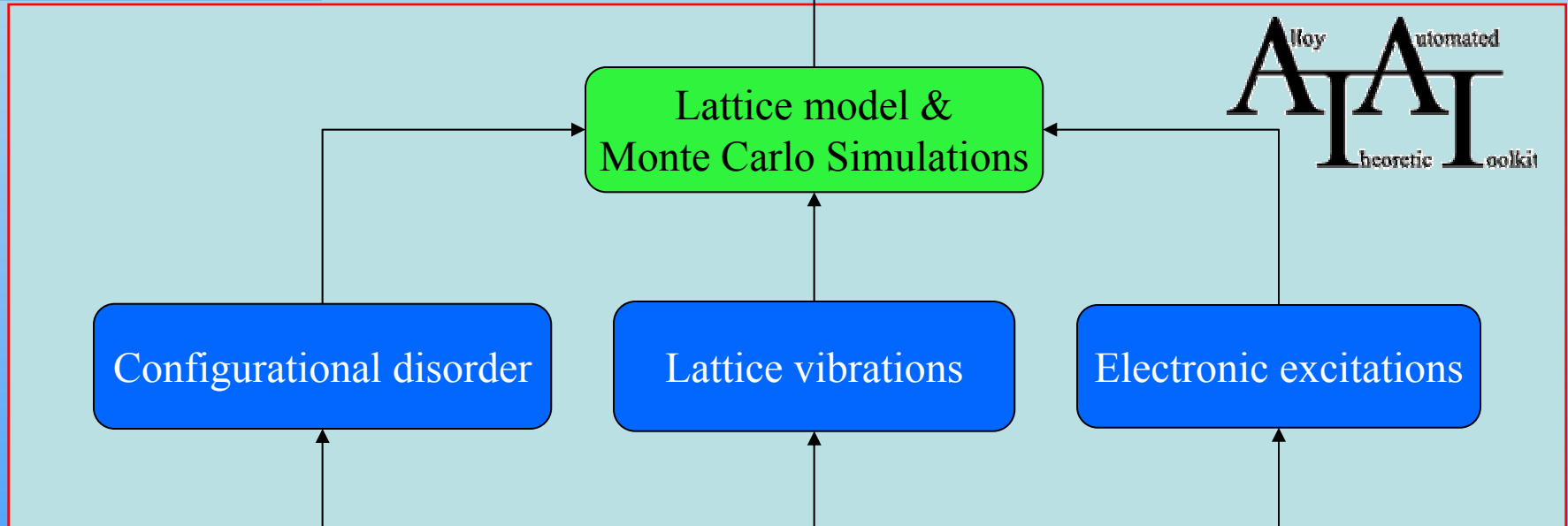
- Example of *ab initio* calculations
- Why *ab initio* Thermodynamics?
- **Methods**
- Applications

# First-principles Thermodynamic Calculations



- Large number of atoms
- Many configurations

Thermodynamic data



Quantum Mechanical Calculations

- Small number of atoms
- Few configurations

Thermodynamic data

Lattice model &  
Monte Carlo Simulations

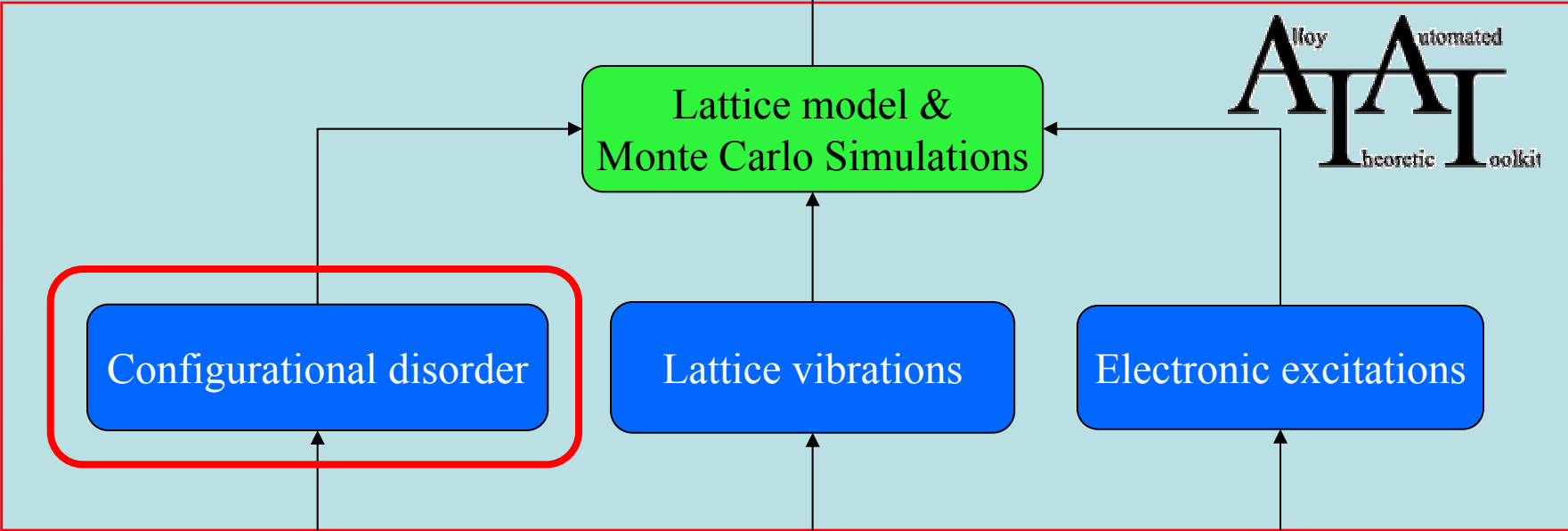
Alloy Automated  
Theoretic Toolkit

Configurational disorder

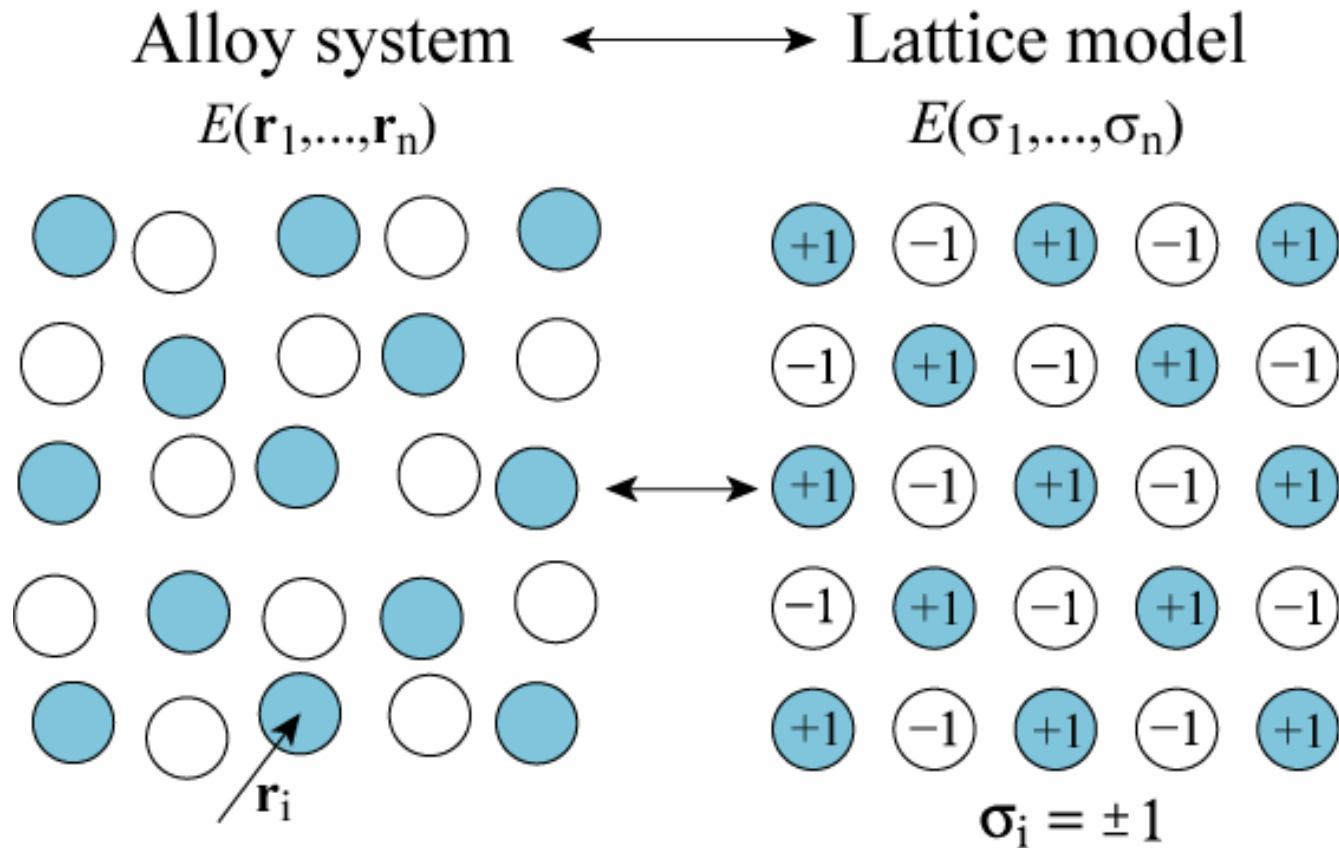
Lattice vibrations

Electronic excitations

Quantum Mechanical Calculations



# The Cluster Expansion Formalism



$$E(\sigma_1, \dots, \sigma_n) = \sum_{\{i,j\}} J_{ij} \sigma_i \sigma_j + \sum_{\{i,j,k\}} J_{ijk} \sigma_i \sigma_j \sigma_k + \dots$$

$$= \sum_{\alpha} J_{\alpha} \sigma_{\alpha}$$

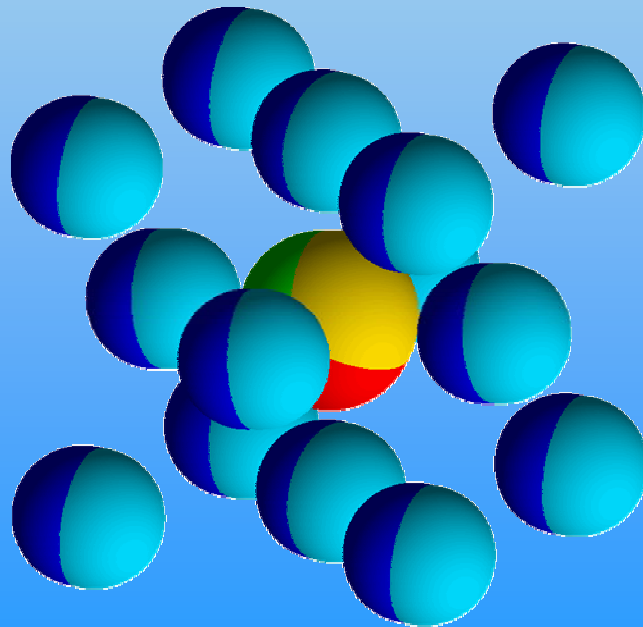
Sanchez, Ducastelle and Gratias (1984)

# Coupled Sublattices Multicomponent Cluster Expansion

Tepesch, Garbulski and Ceder (1995) Sanchez, Ducastelle and Gratias (1984)

$$\text{Same basic form: } E(\sigma_1, \dots, \sigma_n) = \sum_{\alpha} J_{\alpha} \sigma_{\alpha}$$

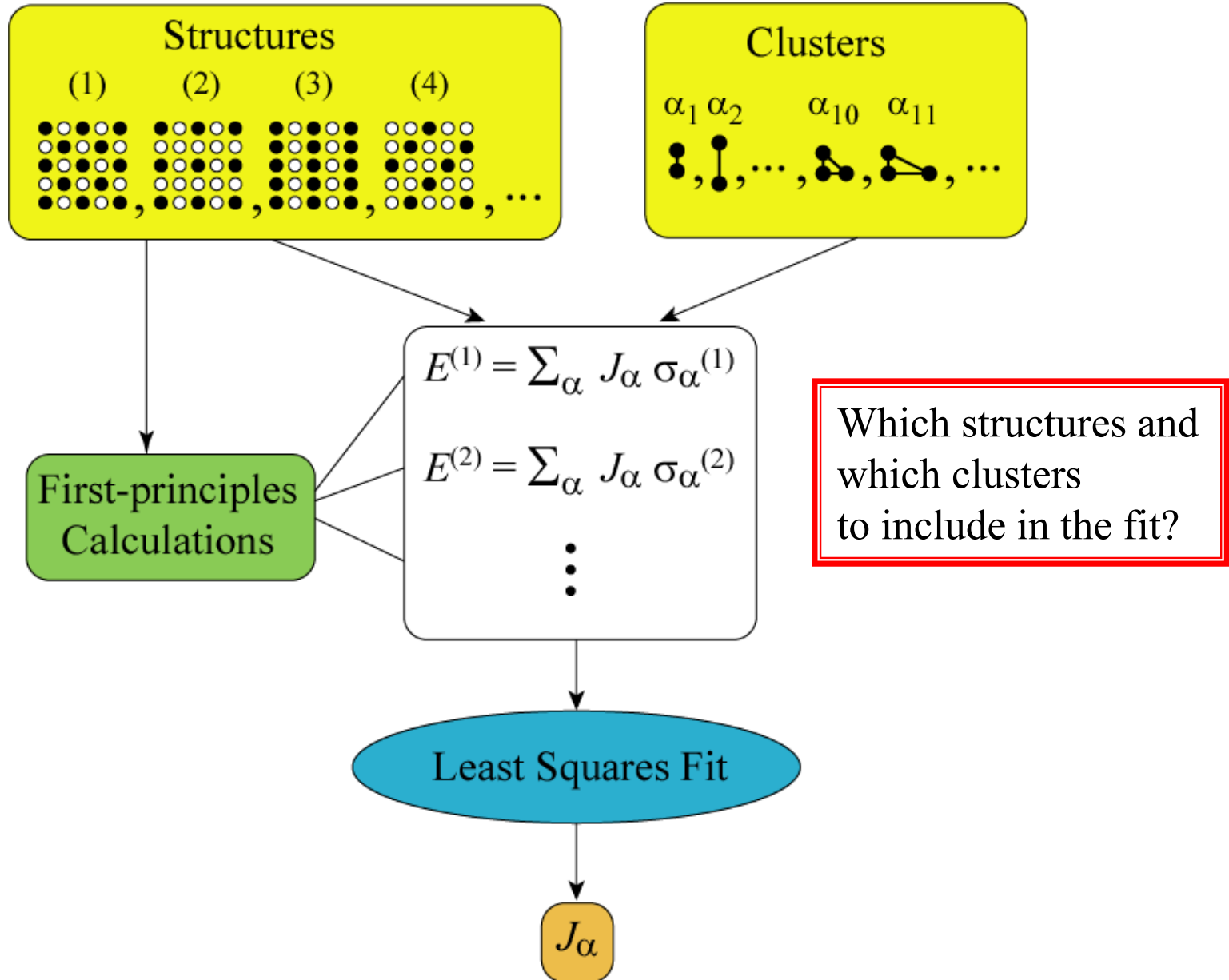
Occupation variables:  $\sigma_i = 0, \dots, n_i - 1$



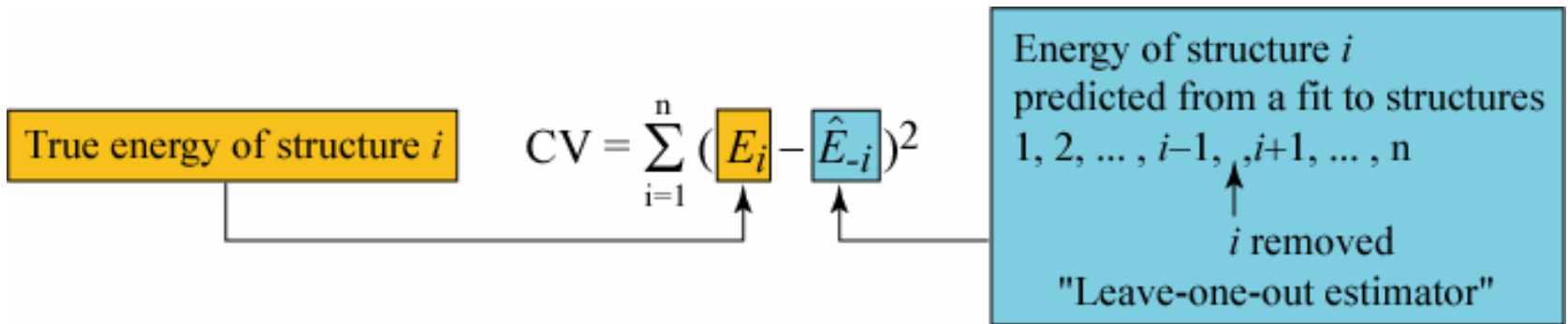
Example: binary fcc sublattice with  
ternary octahedral sites sublattice



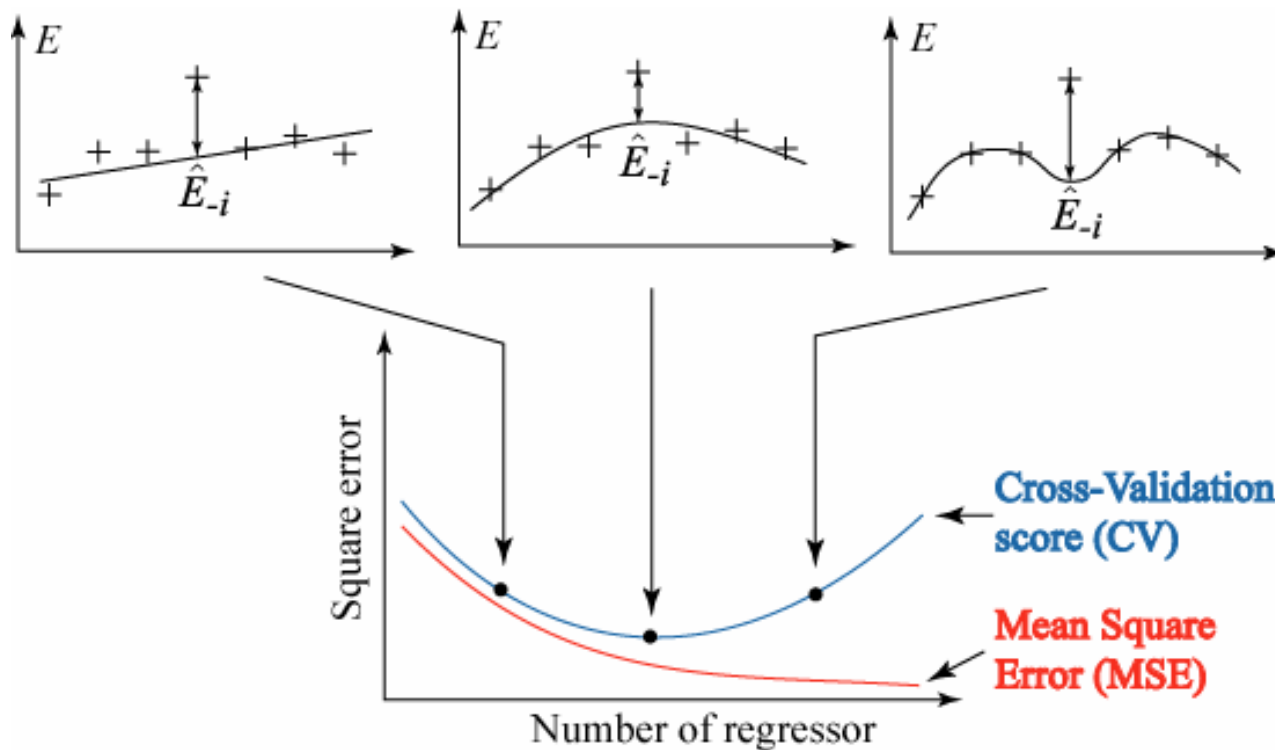
# Cluster expansion fit



# Cross-validation

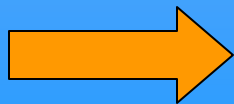
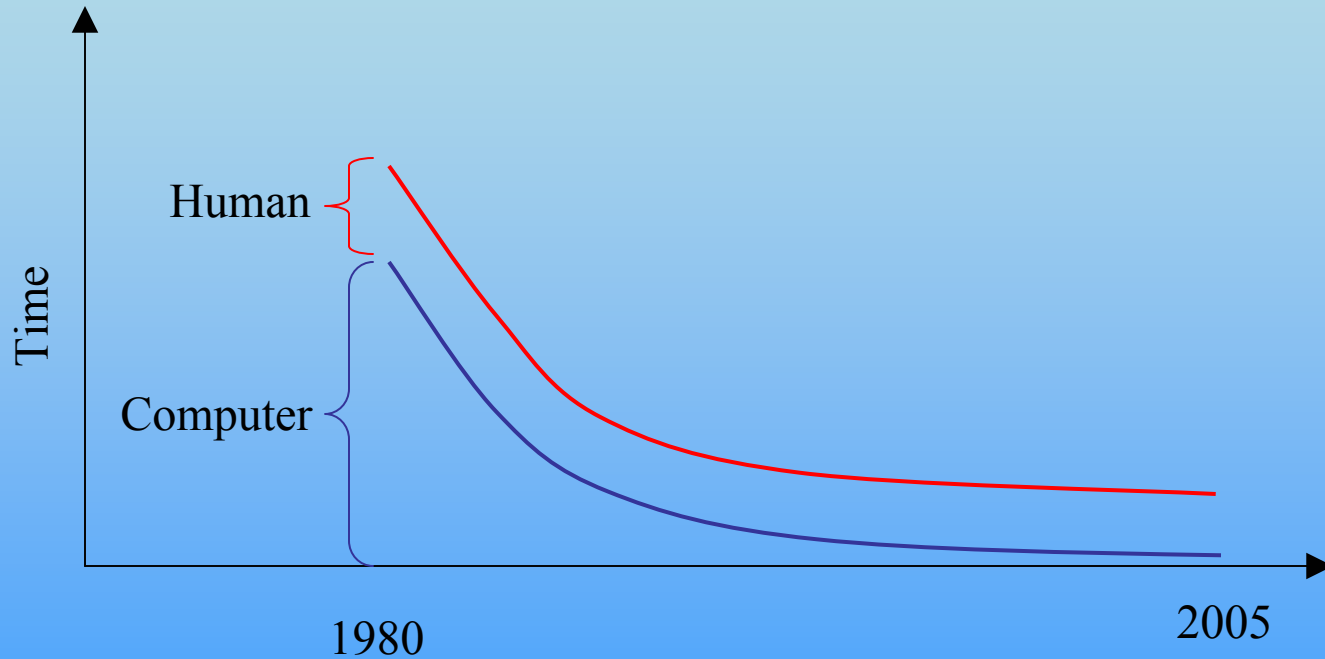


Example of polynomial fit:



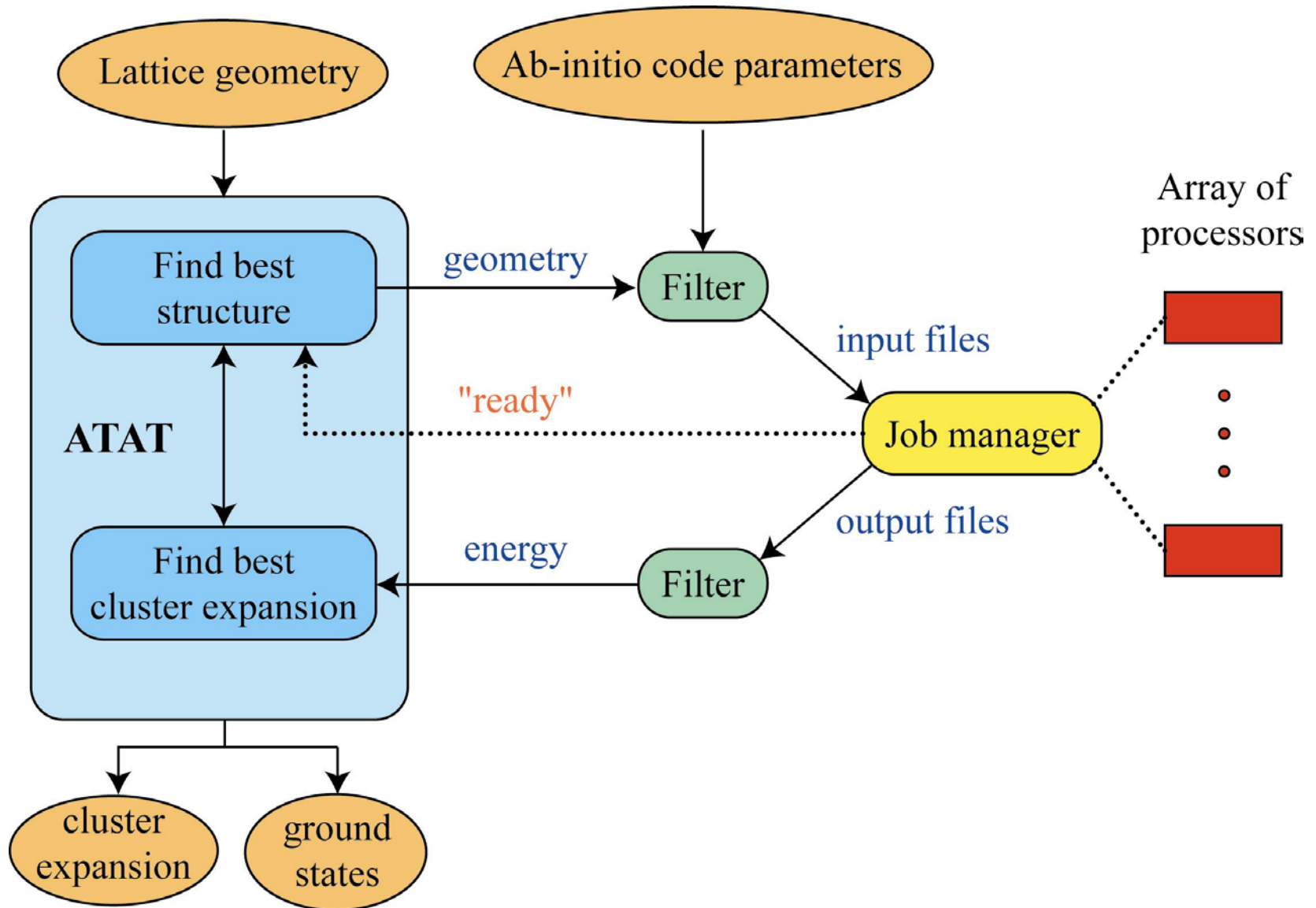
# A matter of time...

Time needed to complete a given first-principles calculation



The procedure needs to be automated

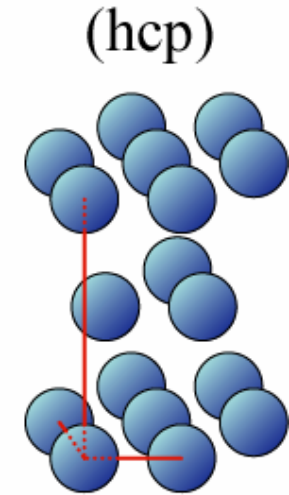
# Automated Cluster Expansion Construction



# Application to Ti-Al Alloys

Simple lattice input file

	a	b	c	$\alpha$	$\beta$	$\gamma$	
Coordinate system	2.94	2.94	4.8	90	90	120	
Unit cell	1	0	0				
	0	1	0				
	0	0	1				
Atoms	0.666667	0.333333	0.500000				Al, Ti
	0.000000	0.000000	0.000000				Al, Ti



Simple ab initio code input file

```
[INCAR]
PREC = high
ISMEAR = -1
SIGMA = 0.1
NSW=41
IBRION = 2
ISIF = 3

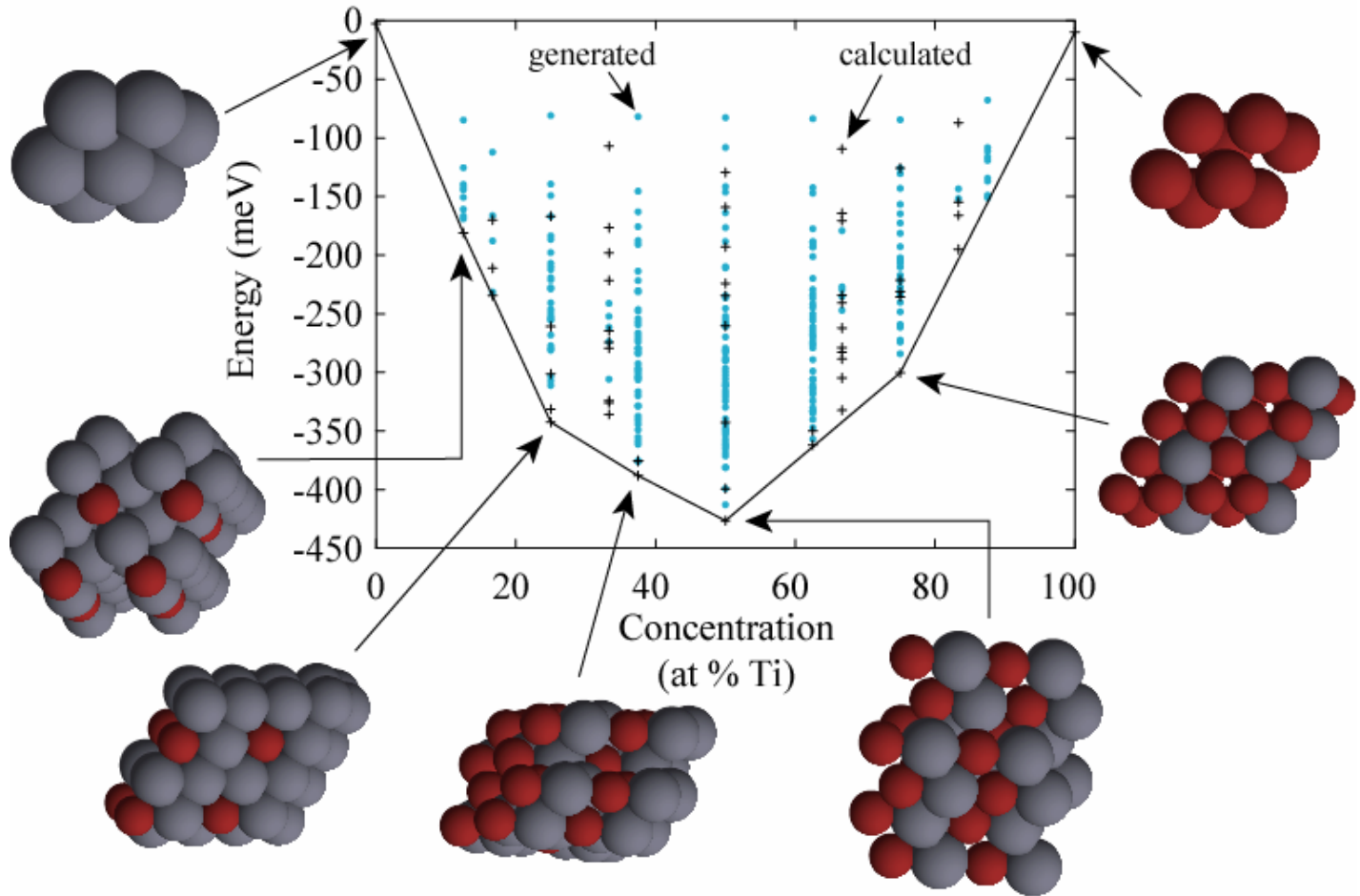
KPPRA = 1000
DOSTATIC
```

} Standard VASP tokens

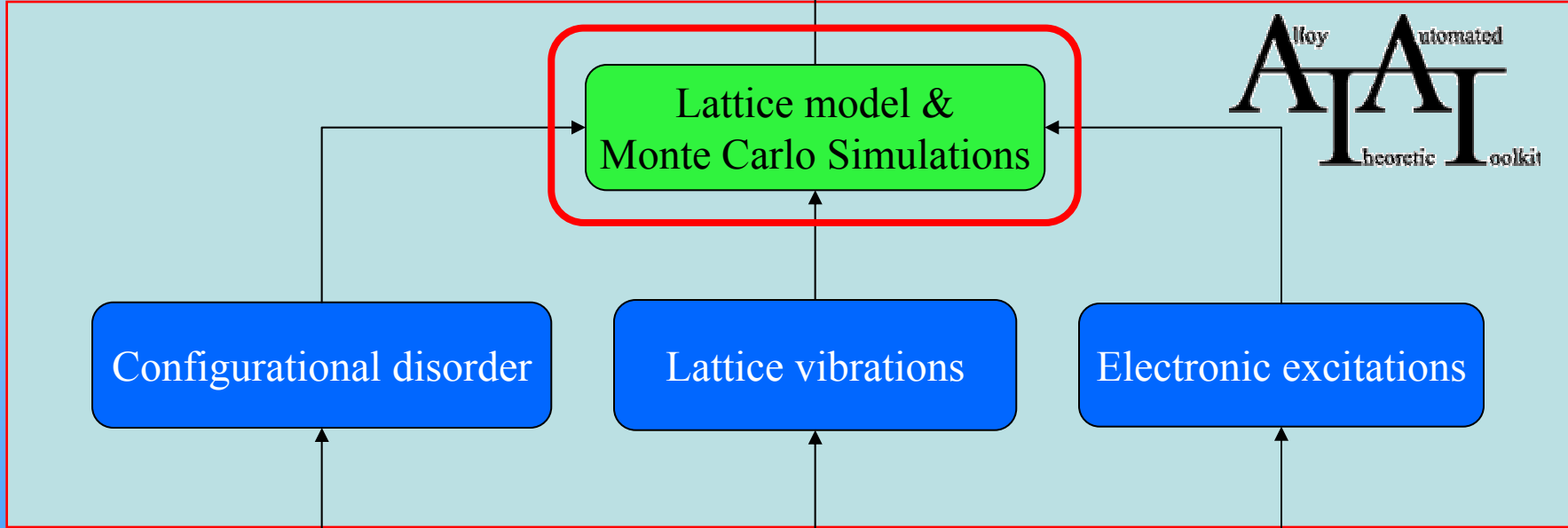
{  $k$ -point density  
{ ( $k$  point per reciprocal atom)

do static run

# Ground States Search



Thermodynamic data



Lattice model & Monte Carlo Simulations

Alloy Automated Theoretic Toolkit

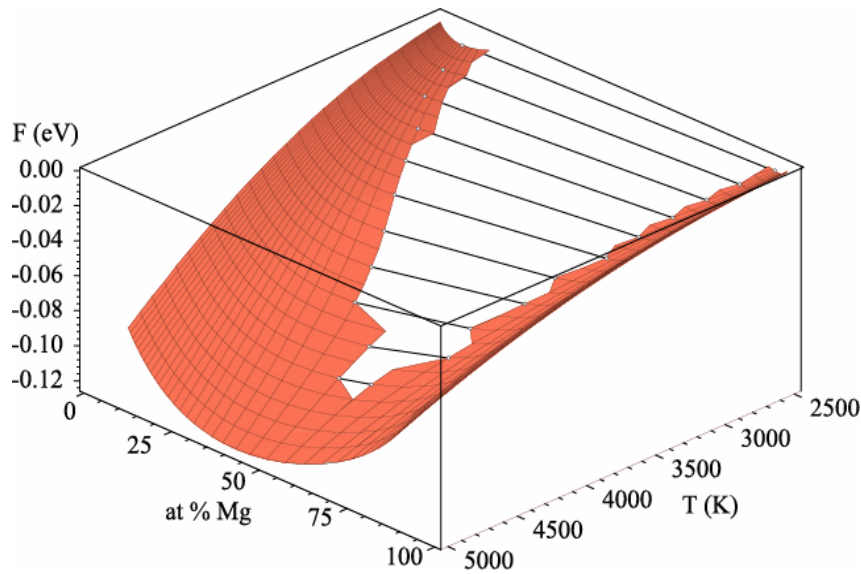
Configurational disorder

Lattice vibrations

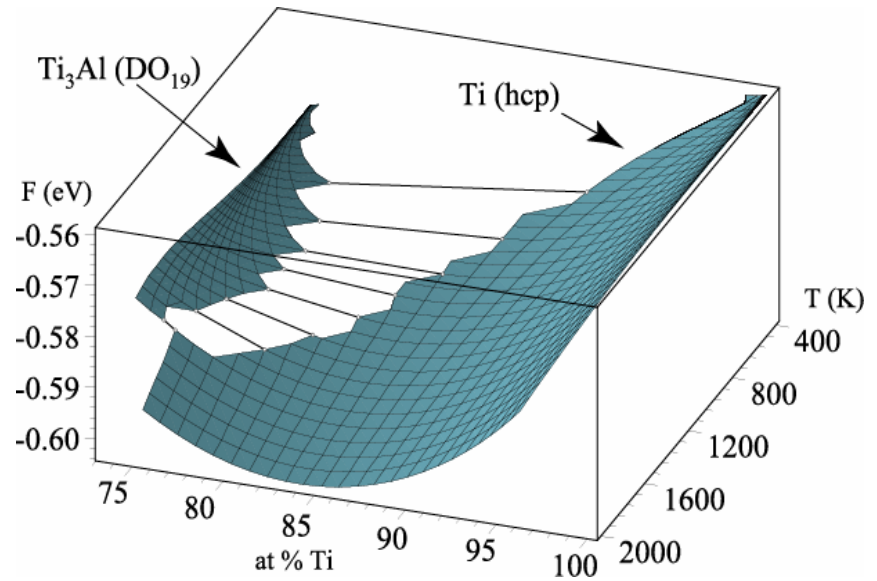
Electronic excitations

Quantum Mechanical Calculations

# Monte Carlo output: Free energies



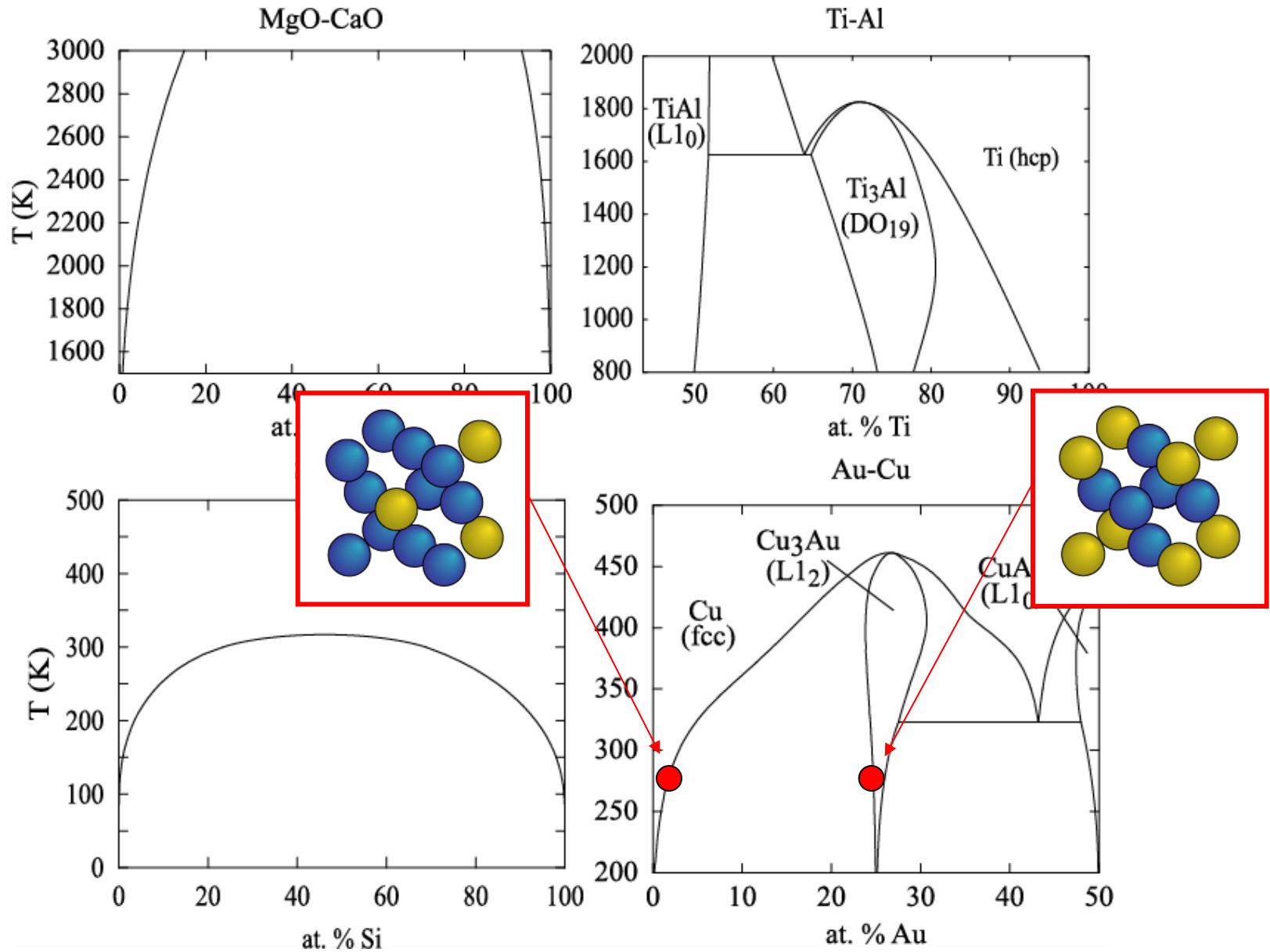
Cao-MgO system



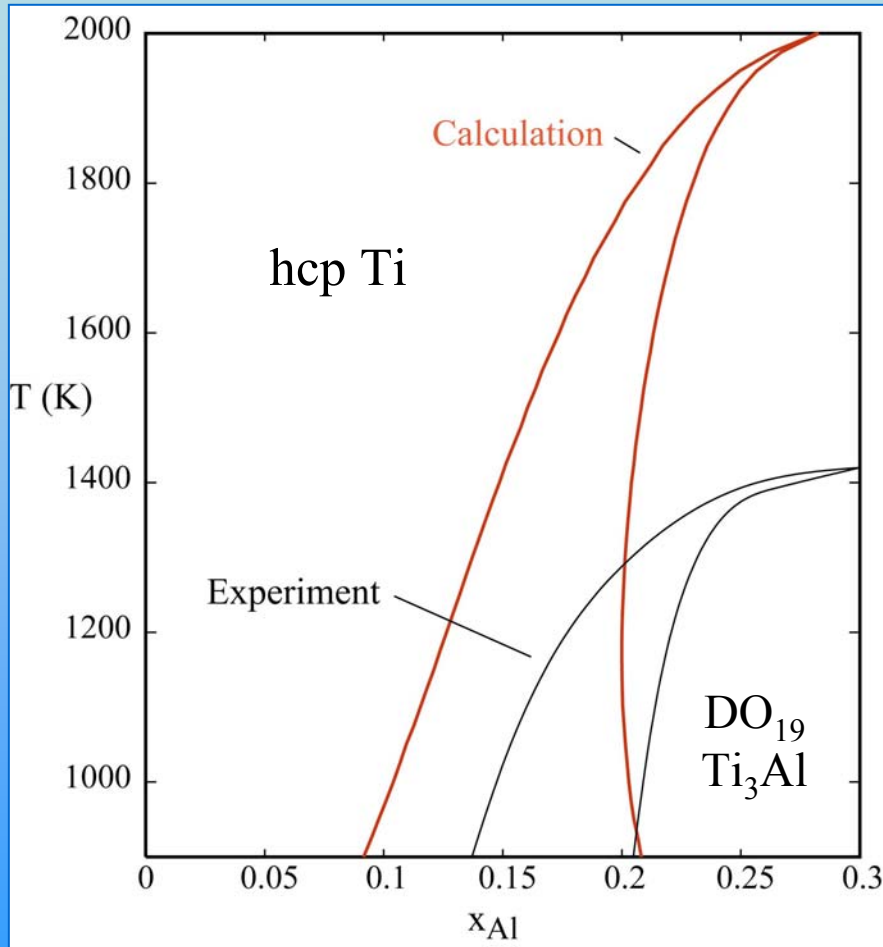
Ti-Al system



# Calculated phase diagram



# Temperature scale problem



Likely source of the discrepancy:  
**Vibrational** entropy.

Fultz, Nagel, Antony, *et al.* (1993-1999)

Ceder, Garbulsky, van de Walle (1994-2002)

de Fontaine, Althoff, Morgan (1997-2000)

Zunger, Ozolins, Wolverton (1998-2001)

Van de Walle, Asta and Ceder (2002),  
Murray (1987) (exp.)

Many other examples...

Thermodynamic data

Lattice model &  
Monte Carlo Simulations

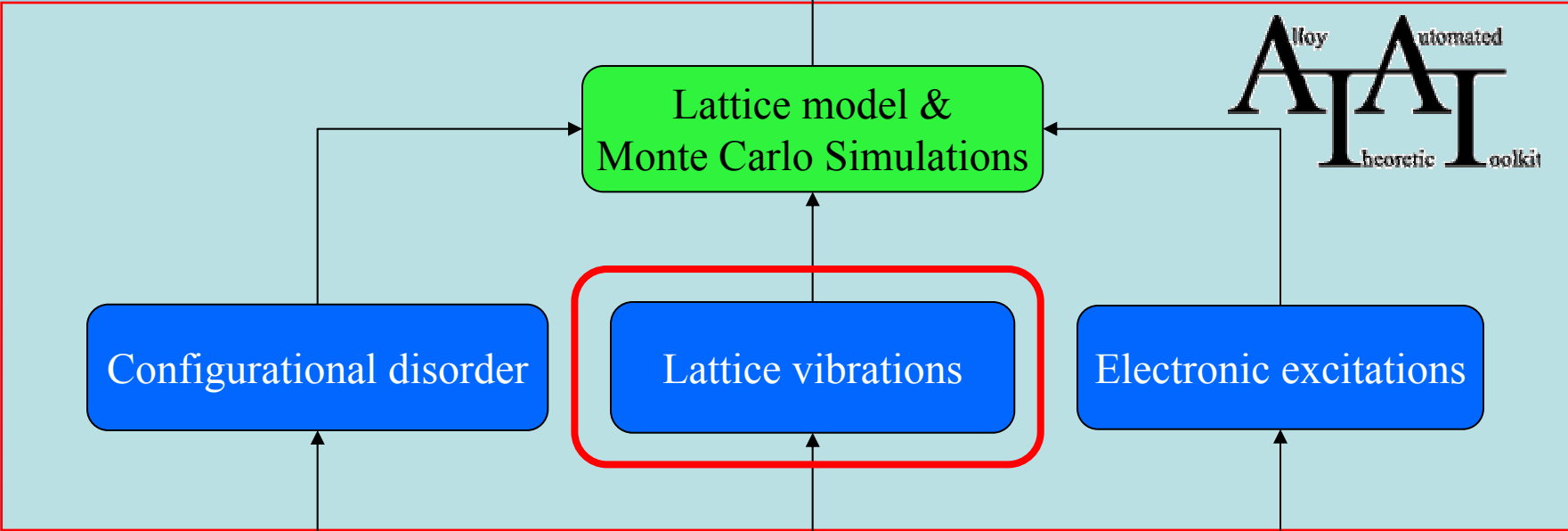
Alloy Automated  
Theoretic Toolkit

Configurational disorder

Lattice vibrations

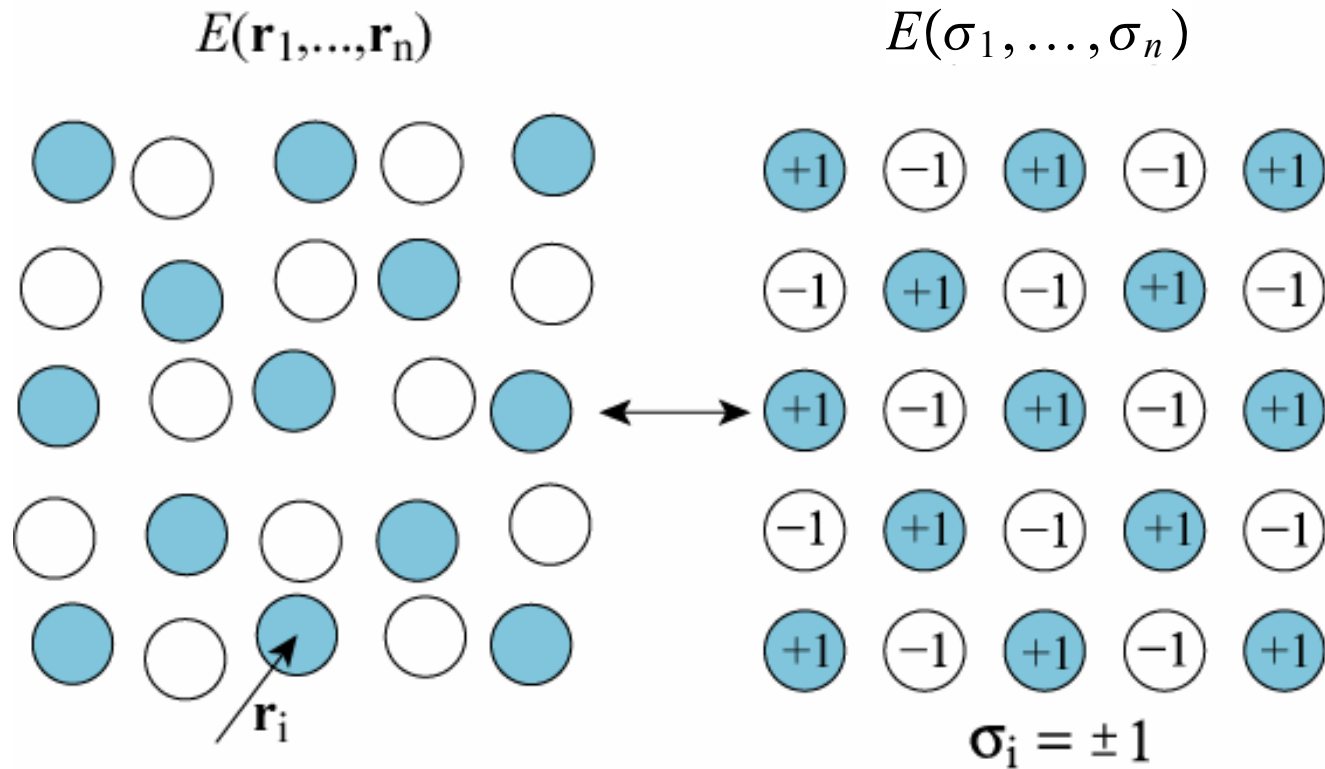
Electronic excitations

Quantum Mechanical Calculations



# The Cluster Expansion Formalism

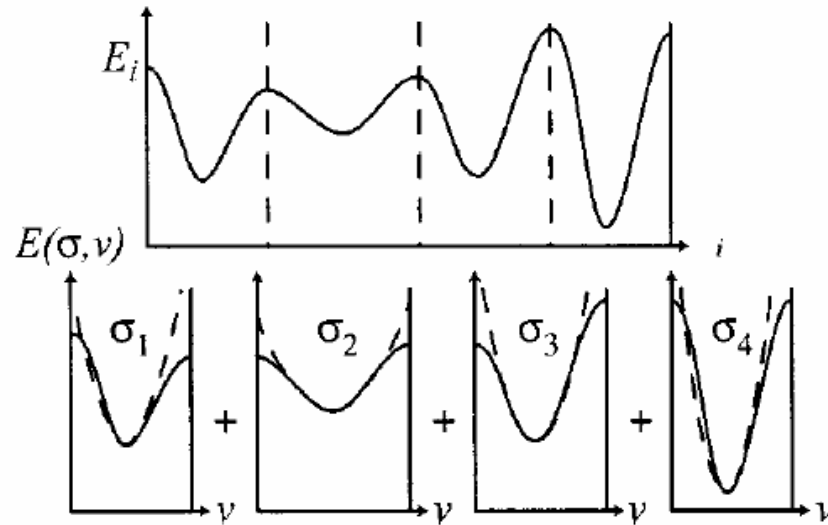
Alloy system  $\longleftrightarrow$  Lattice model



$$F(\sigma_1, \dots, \sigma_n) = \sum_{\alpha} J_{\alpha} \boxed{(T)} \sigma_{\alpha}$$

# Coarse-Graining of the Free Energy

Graphically:



Formally: (Ceder (1993), Garbulski and Ceder (1994-1996))

$$\begin{aligned} F &= -\beta^{-1} \ln\left(\sum_i e^{-\beta E_i}\right) = -\beta^{-1} \ln\left(\sum_{\sigma} \sum_{i \in \sigma} e^{-\beta E_i}\right) \\ &= -\beta^{-1} \ln\left(\sum_{\sigma} e^{-\beta F(\sigma)}\right) \end{aligned}$$

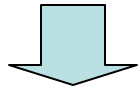
where

$$F(\sigma) = -\beta^{-1} \ln\left(\sum_{i \in \sigma} e^{-\beta E_i}\right)$$

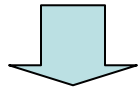
$$\beta = (k_B T)^{-1}$$

# First-principles lattice dynamics

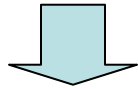
First-principles data



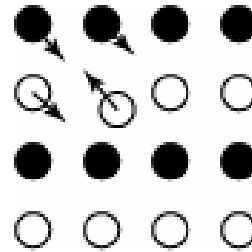
Least-squares fit to Spring model



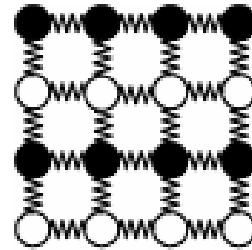
Phonon density of states



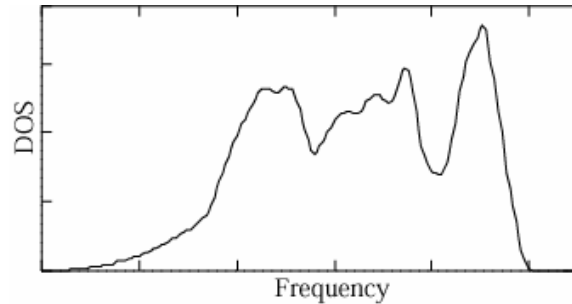
Thermodynamic Properties



Computationally intensive!

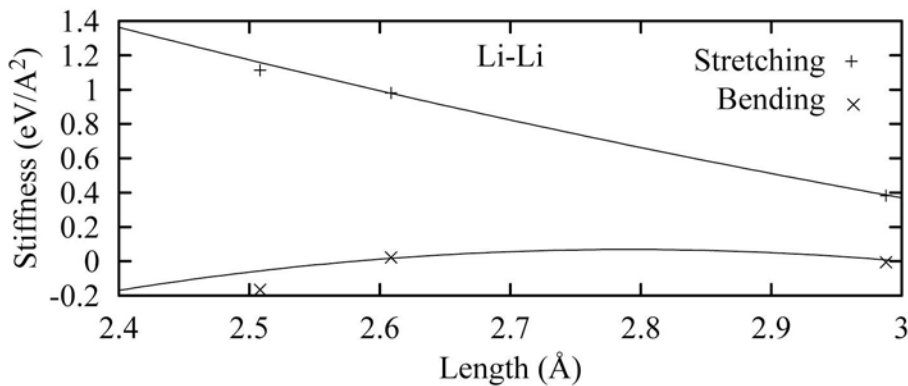
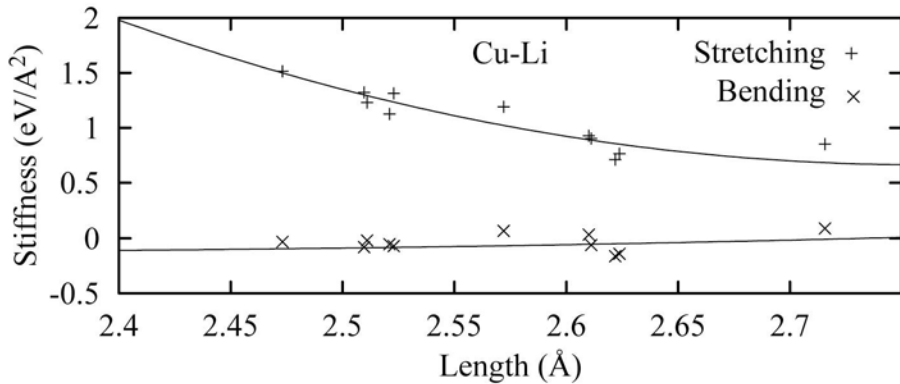
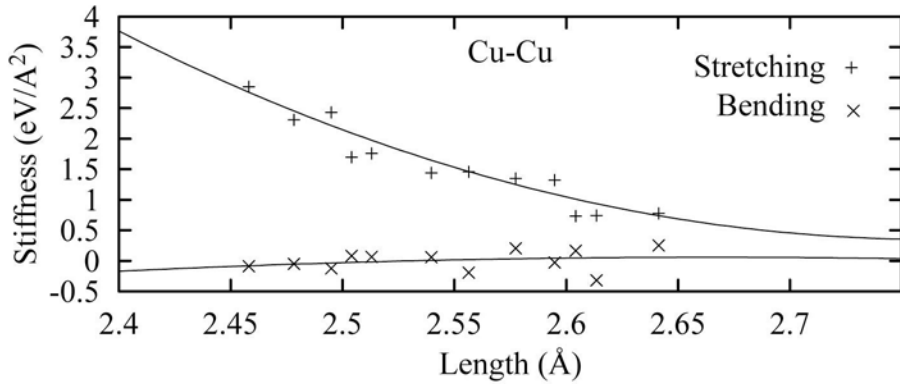


Must be done for many configurations!



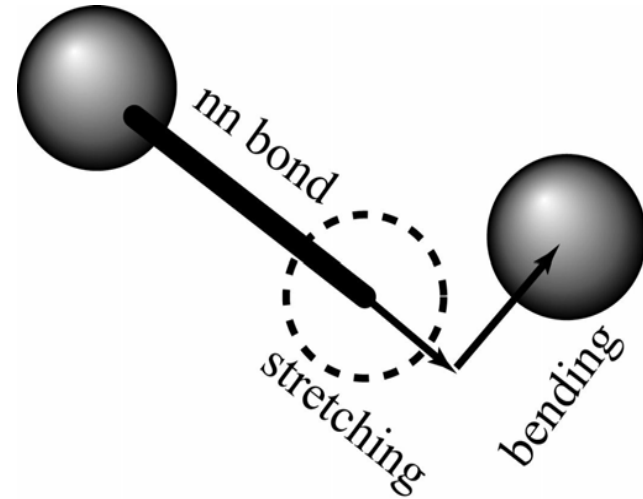
Direct force constant method  
(Wei and Chou (1992), Garbuski and Ceder (1994),  
among many others)

# Transferable Force Constants



Chemical bond type and bond length:

Good predictor of nearest-neighbor force constants (stretching and bending terms)

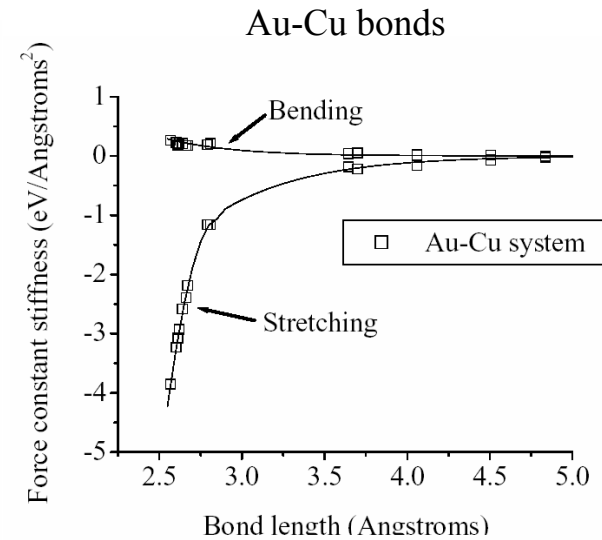
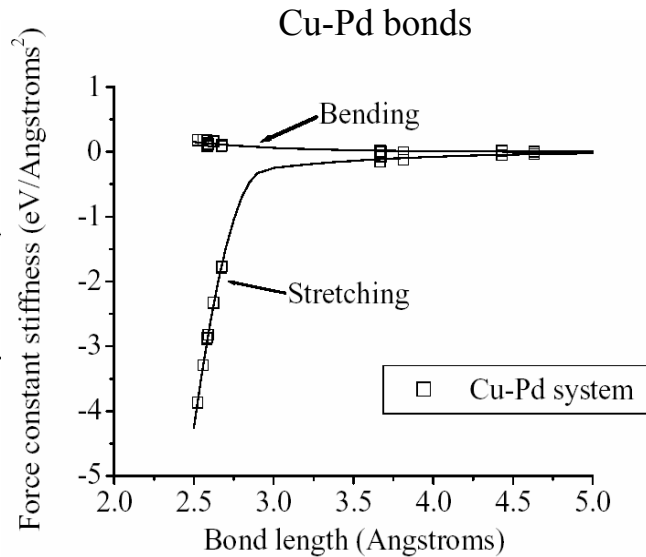
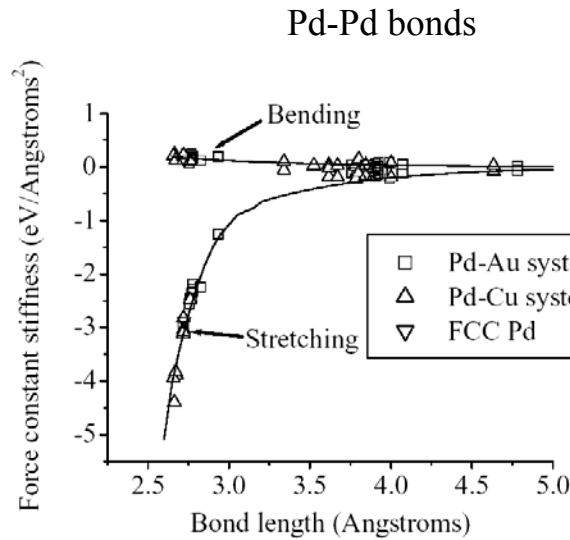
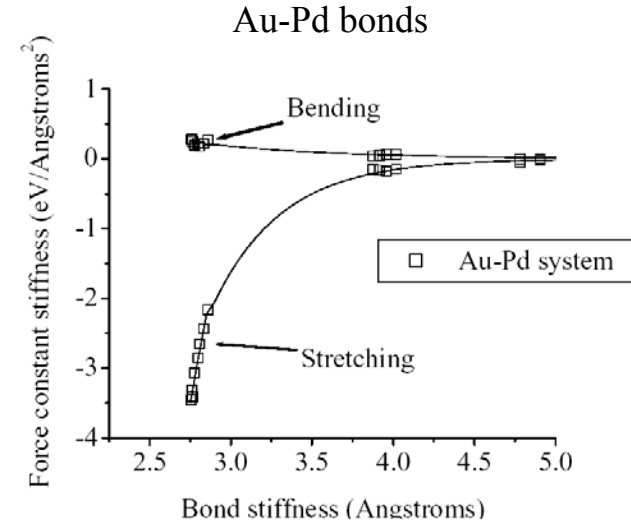
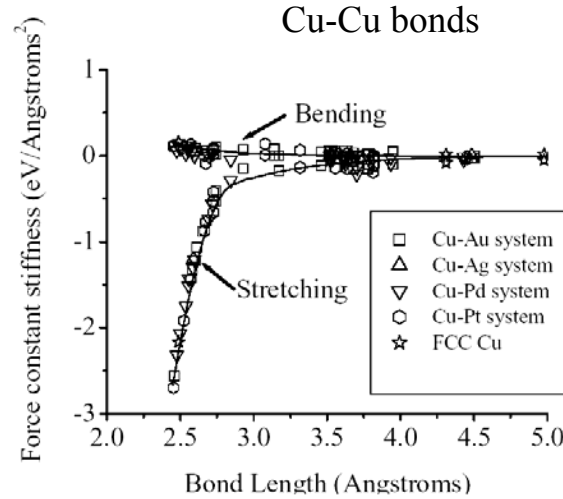
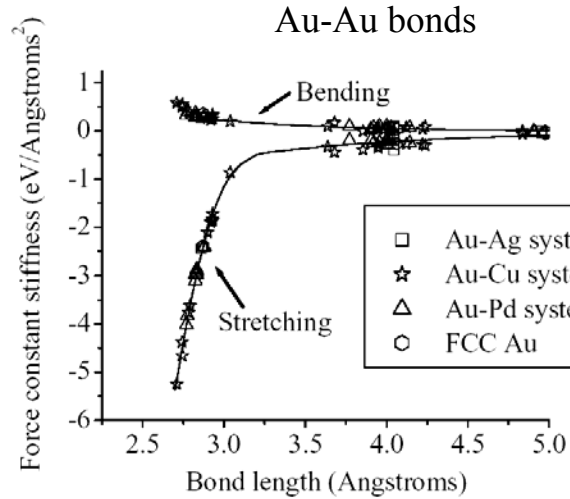


Relationship holds across different structures on the same lattice (here fcc is shown).

van de Walle and Ceder (2000,2002)

# Further tests...

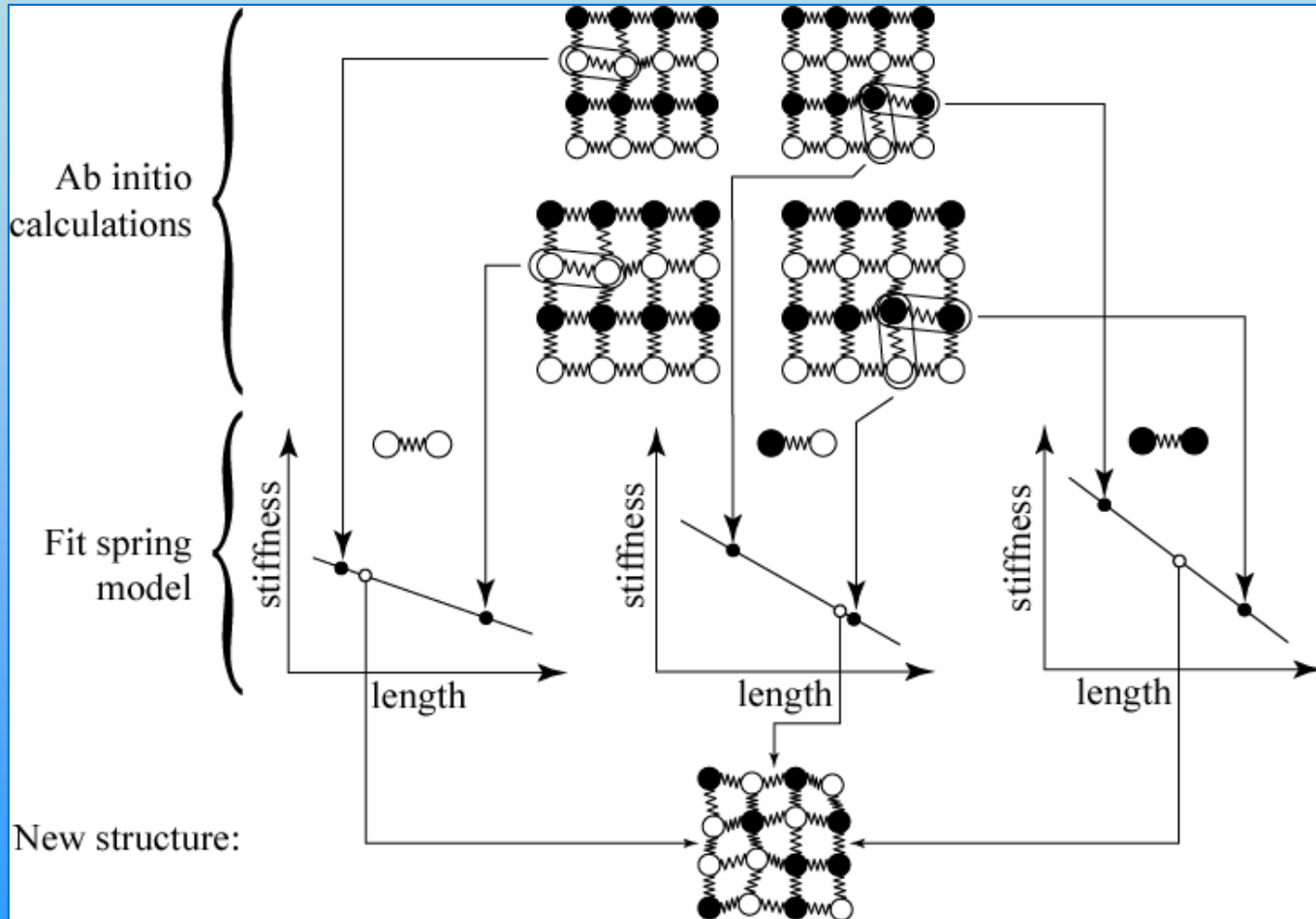
Wu, Ceder, van de Walle (2002)



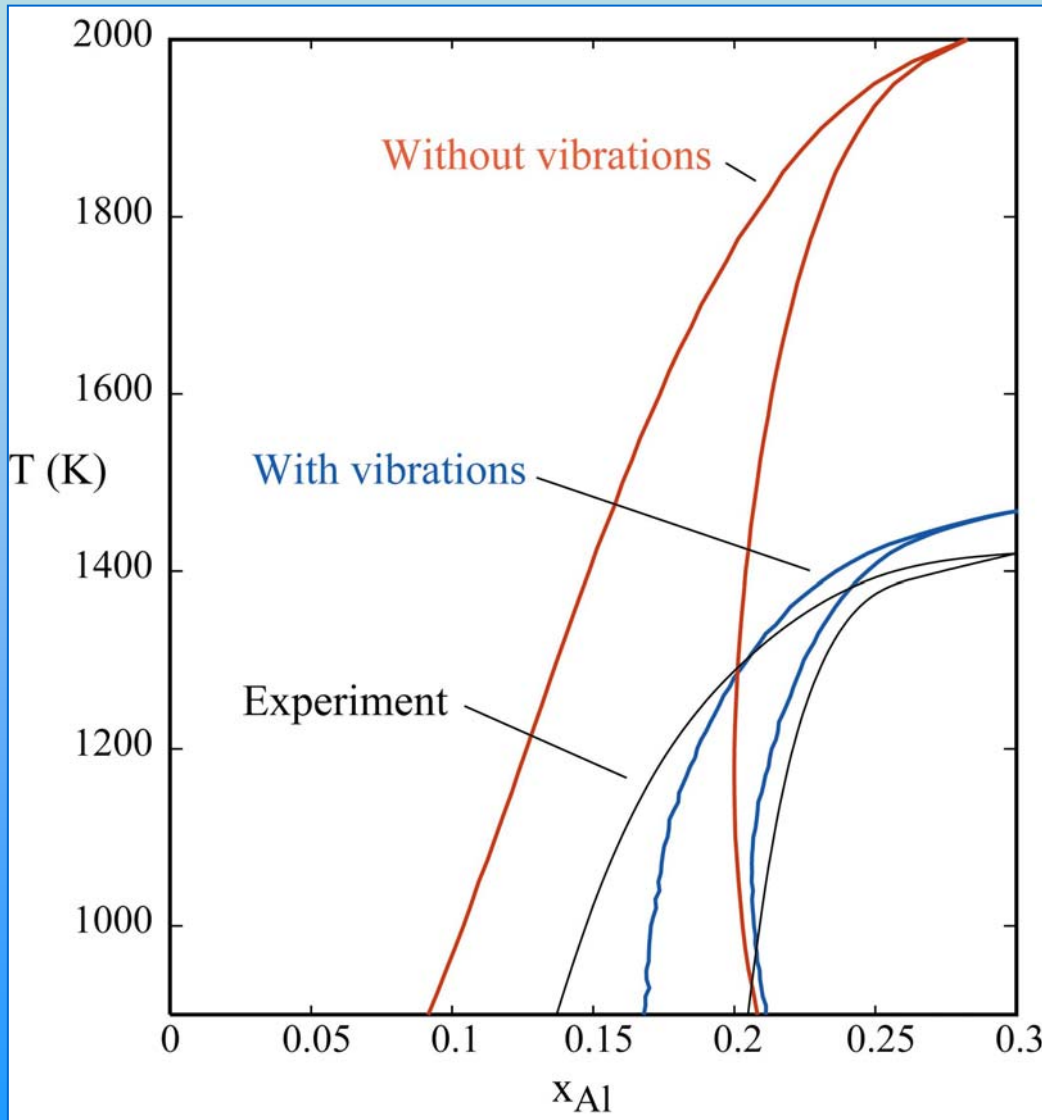
Accuracy  $\sim 0.03 k_B$



# Length-Dependent Transferable Force Constants (LDTFC)



# Calculated Ti-Al Phase Diagram



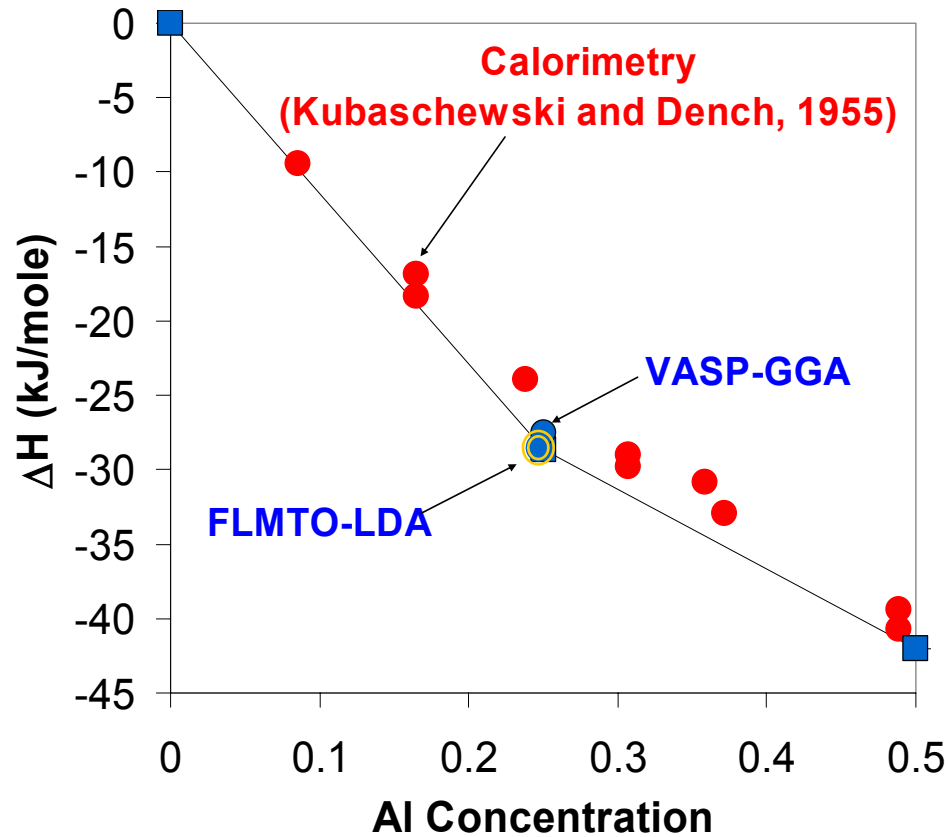
Assessed Phase Diagram:  
I. Ohnuma *et al.*, *Acta Mater.* **48**, 3113 (2000)

1<sup>st</sup>-Principles Calculations:  
van de Walle and Asta

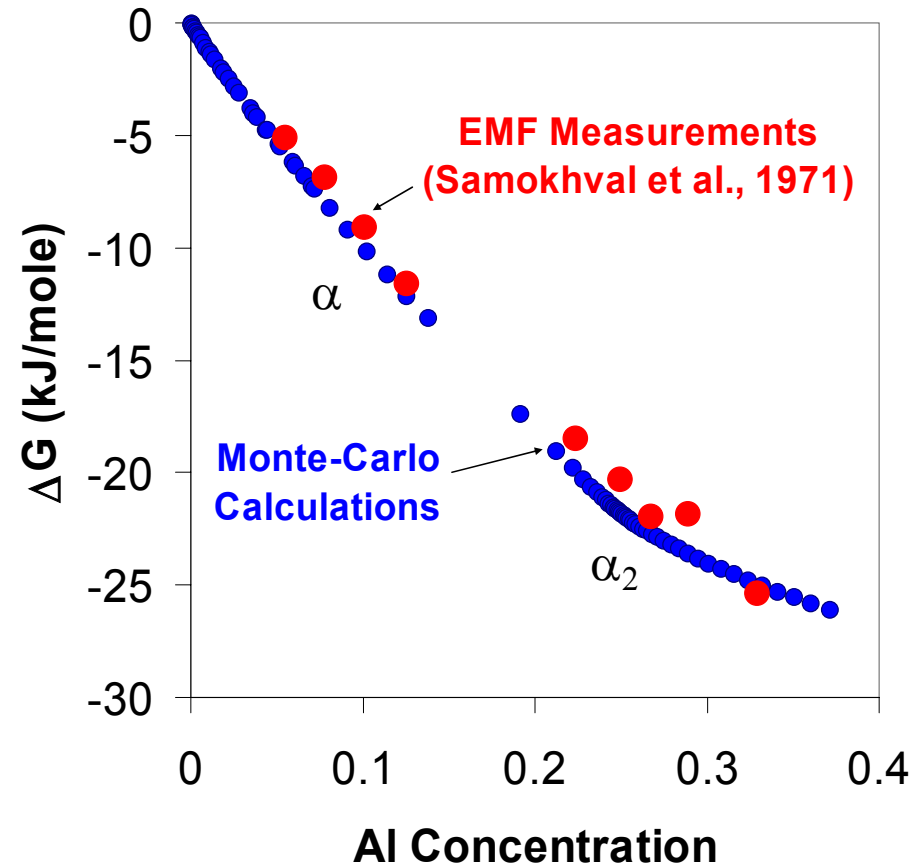
# Ti-Al Thermodynamic Properties

## *1<sup>st</sup>-Principles Calculations vs. Measurements*

### Heats of Formation



### Gibbs Free Energies (T=960 K)



Thermodynamic data

Lattice model &  
Monte Carlo Simulations

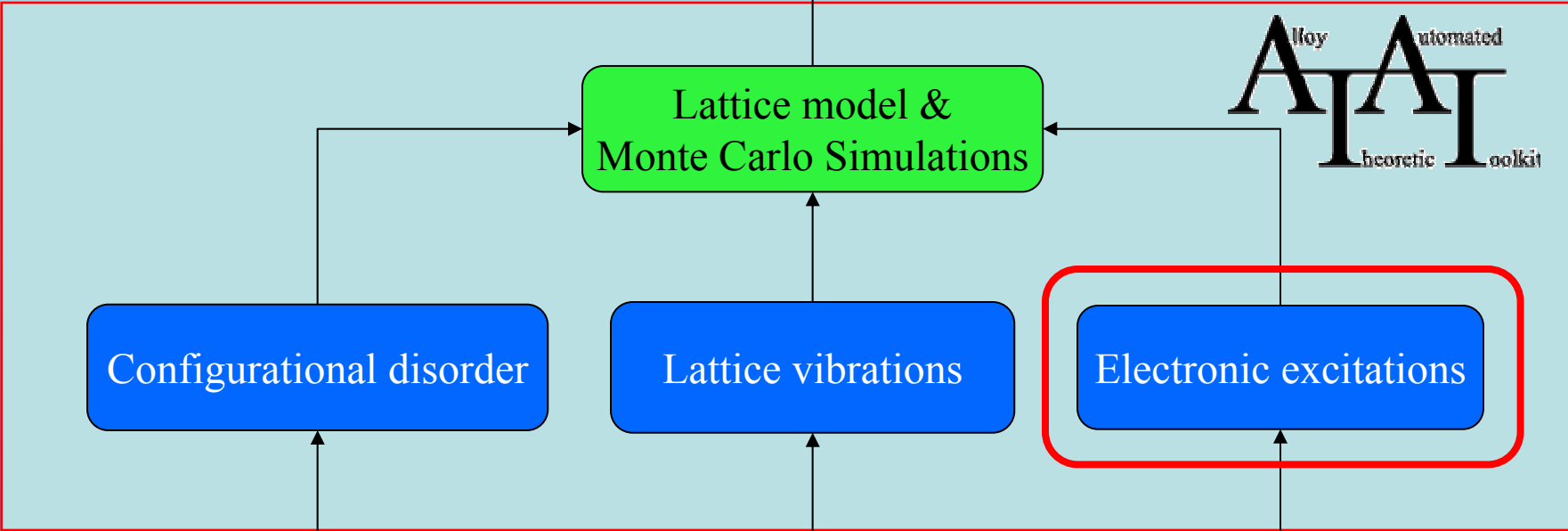
Alloy Automated  
Theoretic Toolkit

Configurational disorder

Lattice vibrations

Electronic excitations

Quantum Mechanical Calculations



# Electronic Excitations

Finite-temperature DFT

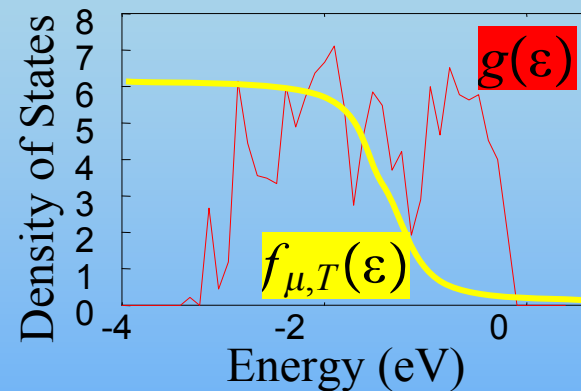
T-independent DOS and charge density

Electronic DOS

Fermi-Dirac Distribution

Electronic Free energy

Advantage: Get  $F_{\text{elec}}$  “for free”



$$F_{\text{elec}}(T) = E_{\text{elec}}(T) - E_{\text{elec}}(0) - TS_{\text{elec}}(T)$$

$$E_{\text{elec}}(T) = \int f_{\mu,T}(\epsilon) \epsilon g(\epsilon) d\epsilon$$

$$S_{\text{elec}}(T) = -k_B \int (f_{\mu,T}(\epsilon) \ln f_{\mu,T}(\epsilon) + (1 - f_{\mu,T}(\epsilon)) \ln(1 - f_{\mu,T}(\epsilon))) g(\epsilon) d\epsilon$$

Cluster expansion:  $F(\sigma_1, \dots, \sigma_n) = \sum_{\alpha} J_{\alpha}(T) \sigma_{\alpha}$



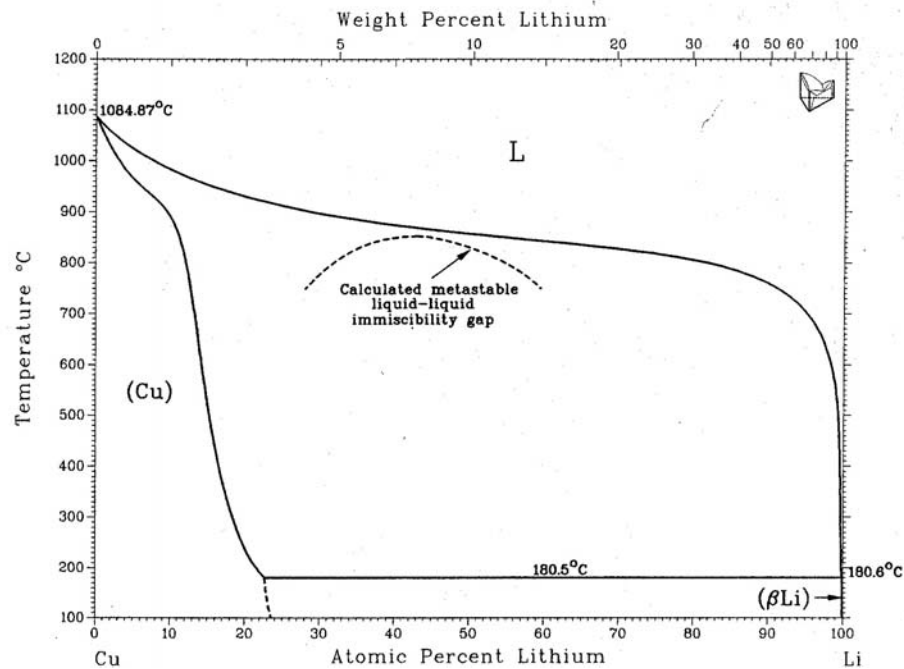
Or: any other model of electronic entropy

# Outline

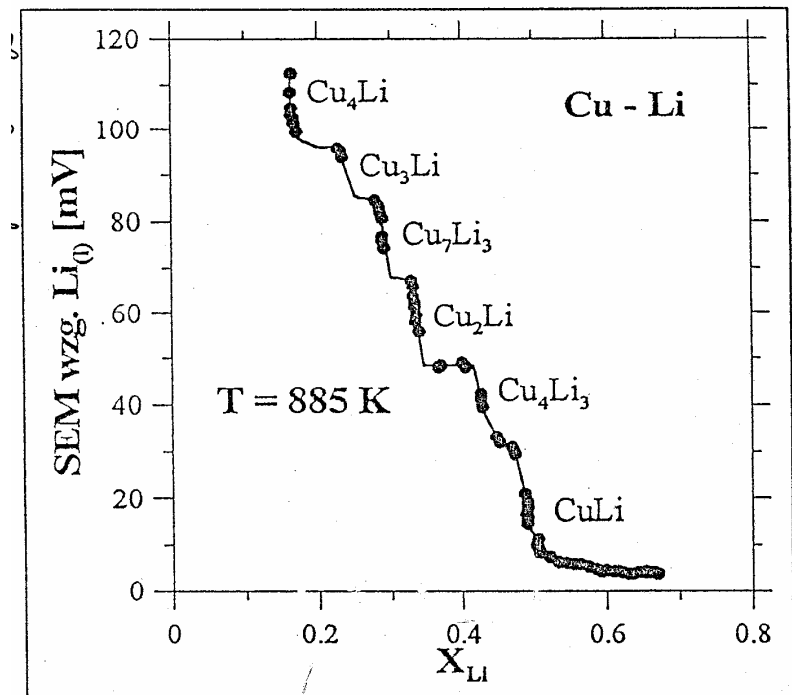
- Example of *ab initio* calculations
- Why *ab initio* Thermodynamics?
- Methods
- Applications

# Ordering in the Cu-Li system?

Widely used assessments  
do not include ordered phases  
(Pelton (1986), Saunders (1998)).

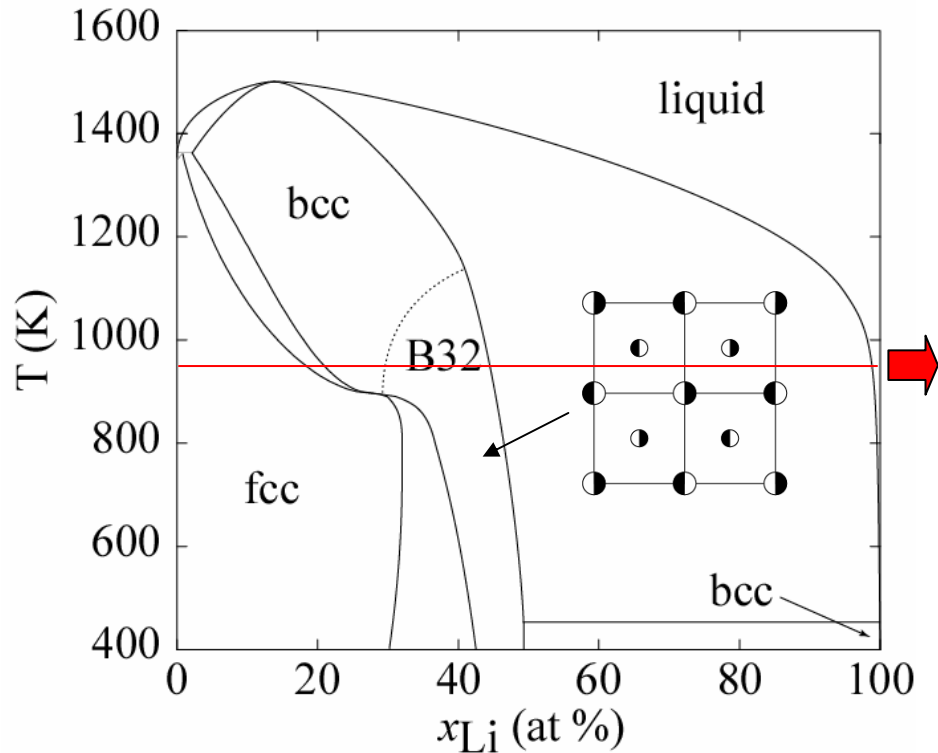


Evidence from EMF measurements  
Ordered  $\text{Cu}_4\text{Li}$  phase also suggested by  
Borgstedt & Gumiński (1996),  
Krauss, Mendelson, Gruen, *et al.* (1986),  
Old & Trawena (1981).

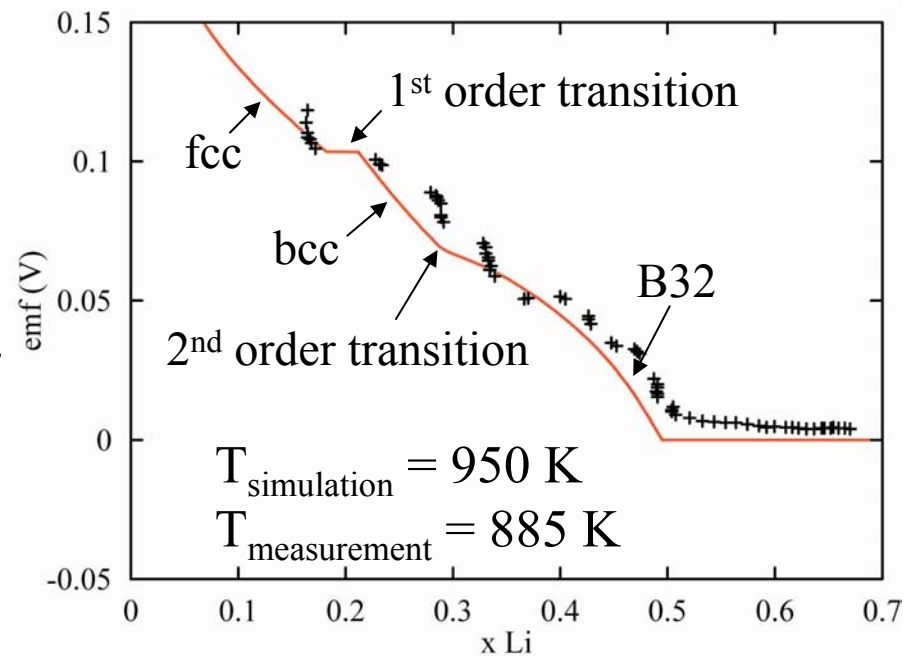


# Calculated Thermodynamic data for Cu-Li system

## Phase diagram



## Electromotive Force (EMF)

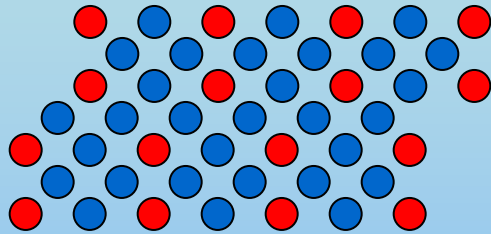


2 adjustable parameters:

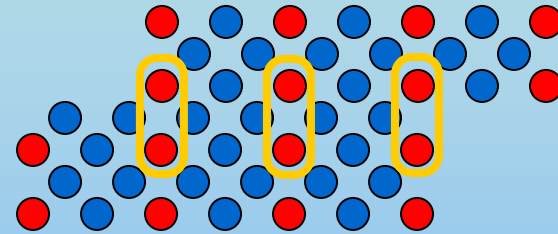
$T_{\text{simulation}}$  + chemical potential of Li (liq)



# Antiphase Boundary

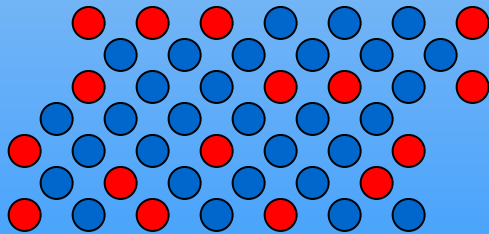


after passage of 1 dislocation

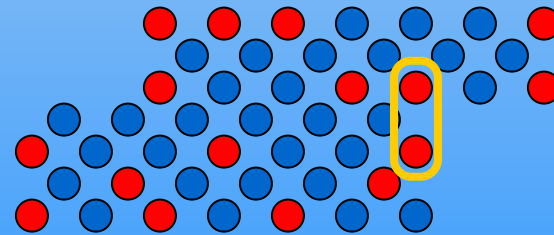


after passage of 2 dislocations

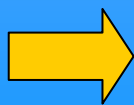
# Diffuse Antiphase Boundary



after passage of 1 dislocation

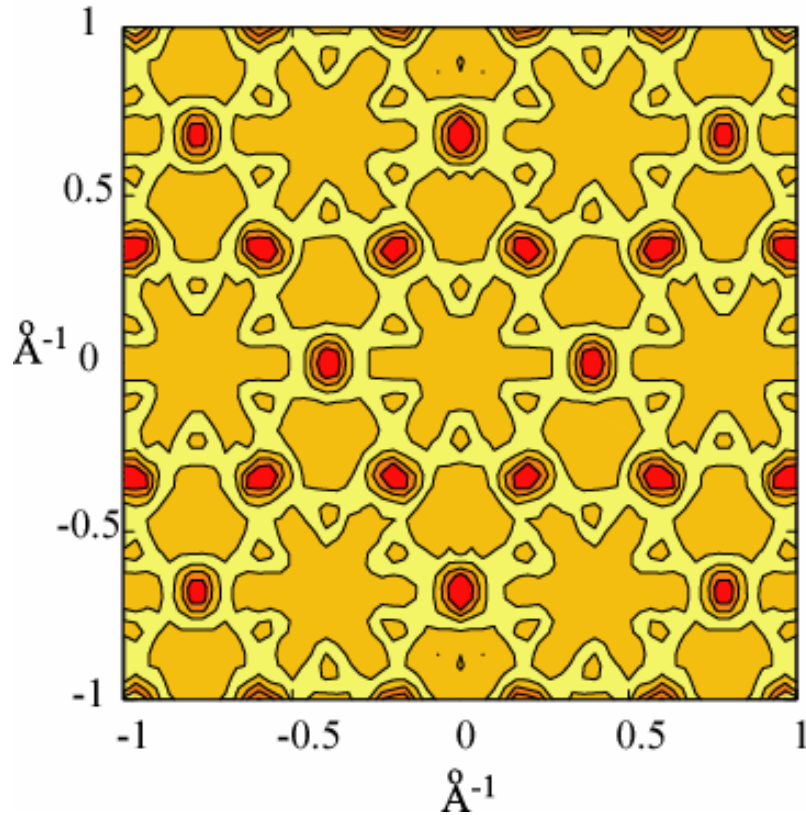


after passage of 2 dislocations

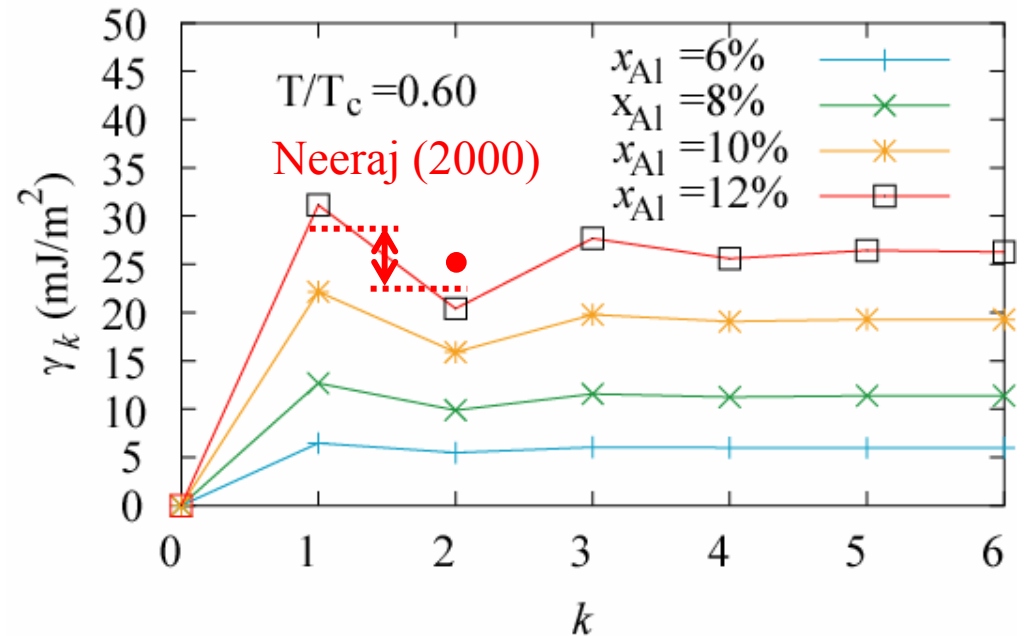


Creation of a plane with easy dislocation motion:  
Work softening

# Short-range order and diffuse antiphase boundary energy calculations

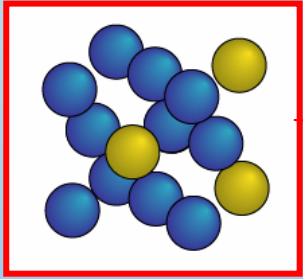


Calculated diffuse X-ray scattering in Ti-Al hcp solid-solution

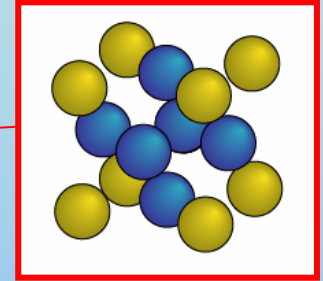
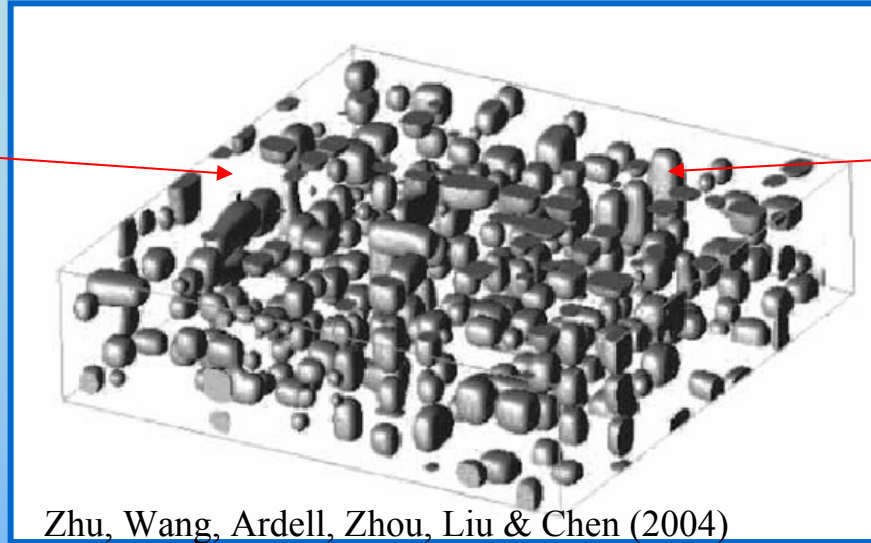


Energy cost of creating a diffuse antiphase boundary in a Ti-Al hcp short-range ordered alloy by sliding  $k$  dislocations

# Phase Field Modeling of System



Ni-rich fcc  
solid solution



$L1_2$   $Ni_3Al$

Free energy:

$$F = \int_V \left[ f(c, \phi_i) + \frac{\varepsilon^2}{2} \sum_i (\nabla \phi_i)^2 \right] dV$$

Thermodynamic term

phase fields

“gradient energy” term

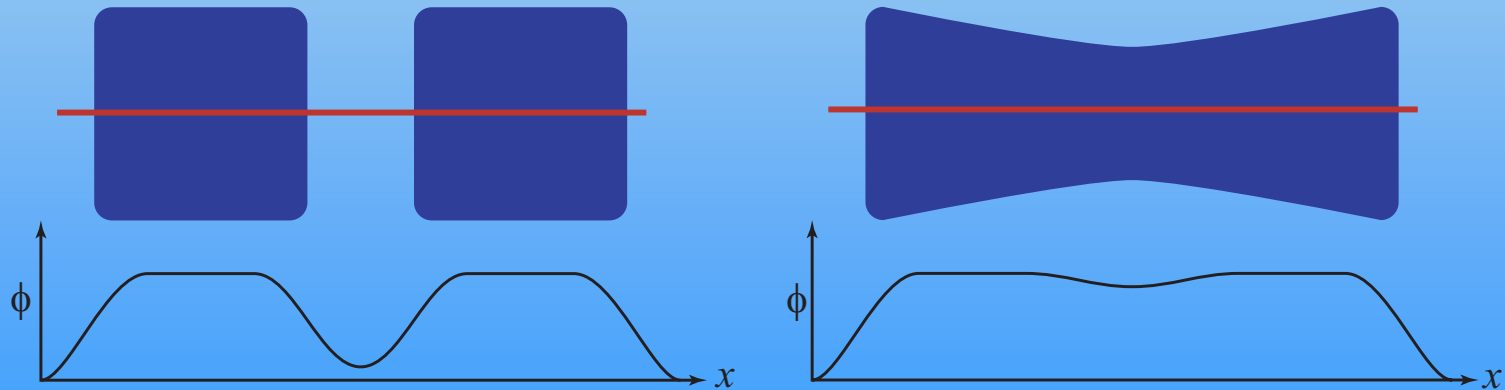
Input parameters needed: Bulk thermodynamic data, Kinetic coefficients,

Interfacial free energy,  
Interface width

# Interface Width & Coalescence

In phase field modeling:

- ◆ Interface width usually a *numerical parameter*.
- ◆ Can get correct behavior with a larger-than-physical width **except** if modeling of coalescence required.



Wider Interface

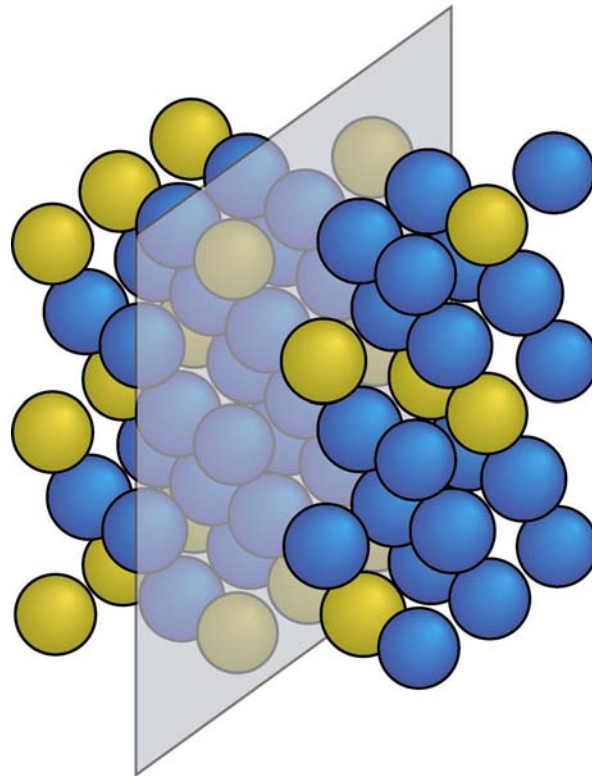
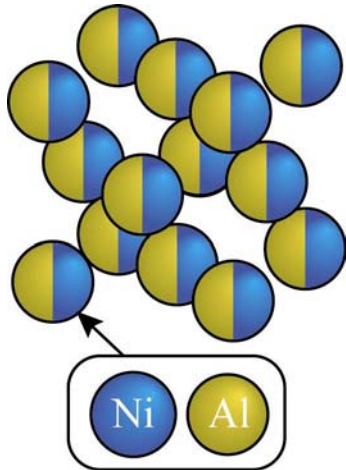


Increased coalescence

# Ni<sub>3</sub>Al/Ni (001)

## Diffuse Coherent Interface

Lattice model

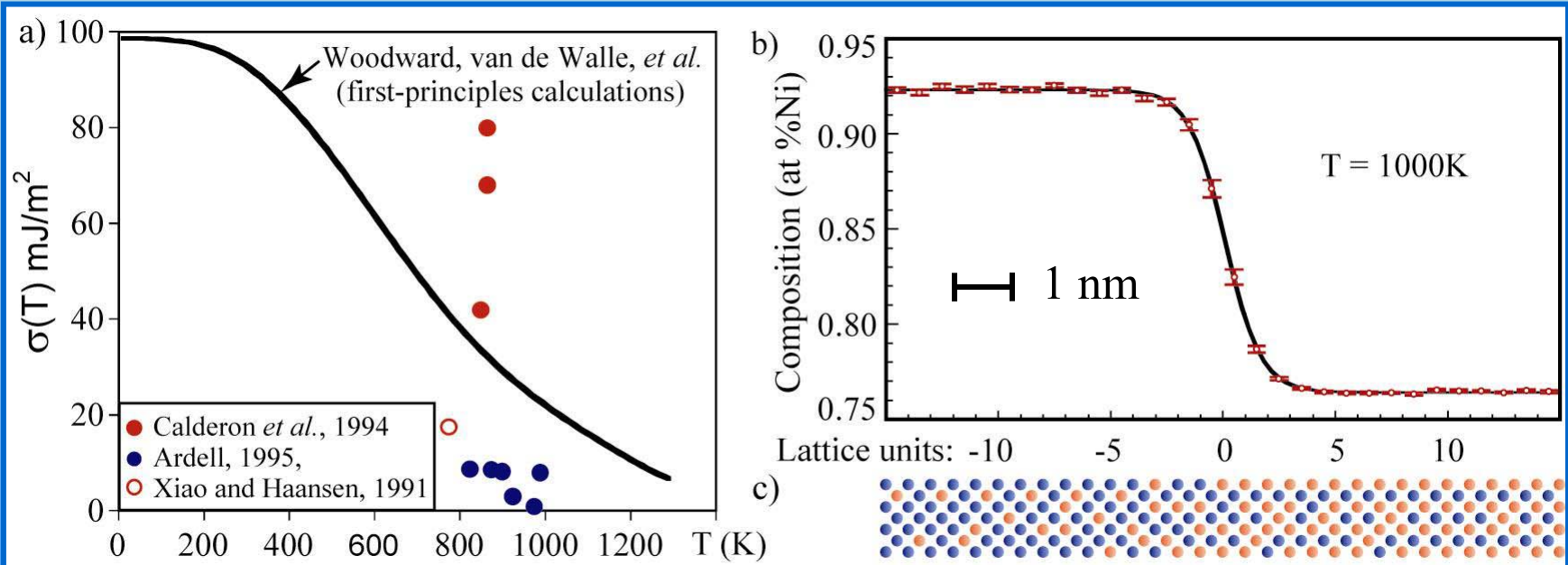


# Interfacial Thermodynamics

Ni<sub>3</sub>Al/Ni (001) diffuse coherent interface

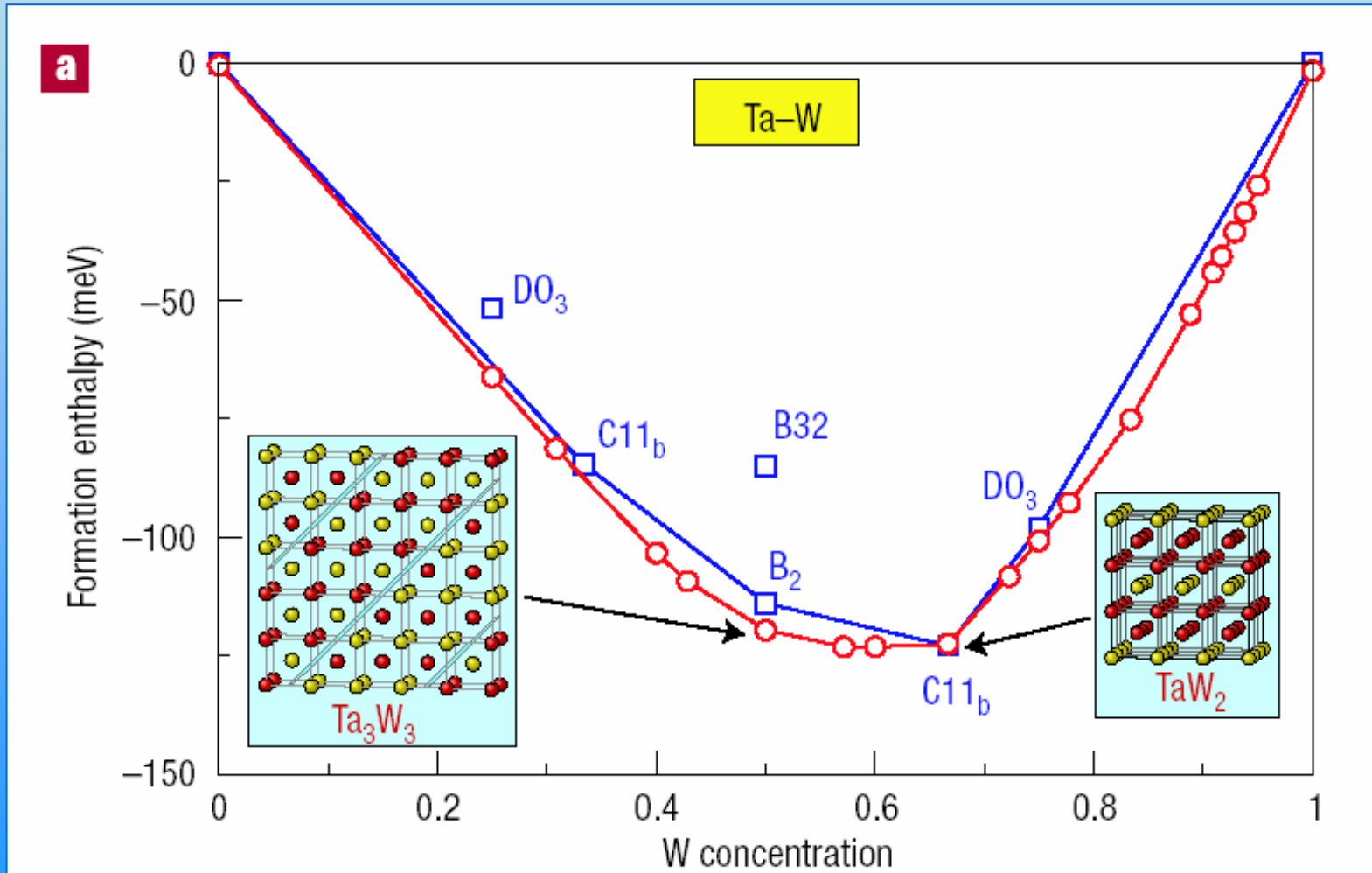
Interfacial free energy

Interface width



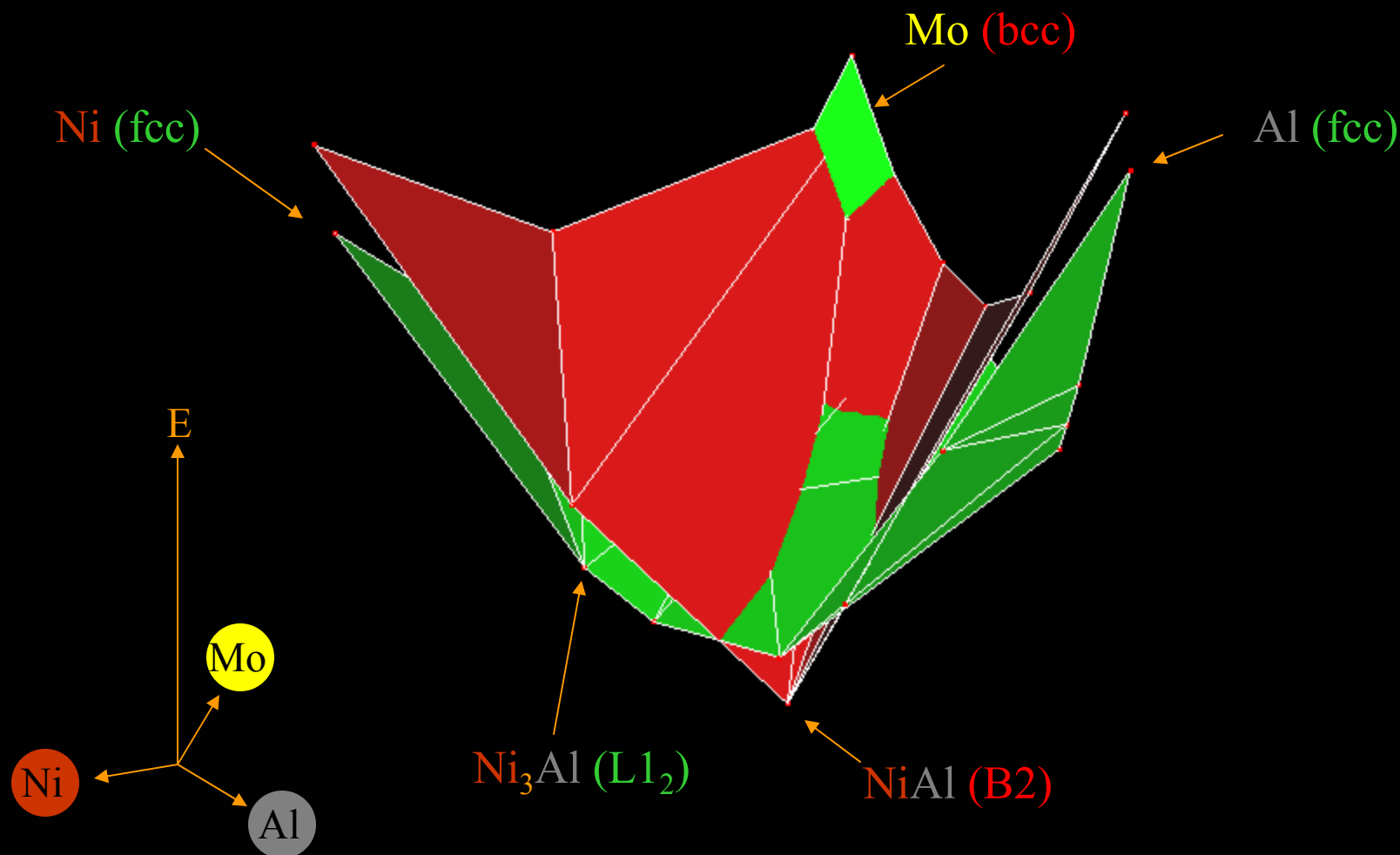
Important input for mesoscale phase field simulations:  
provides the “gradient energy” term

# Search for new compounds



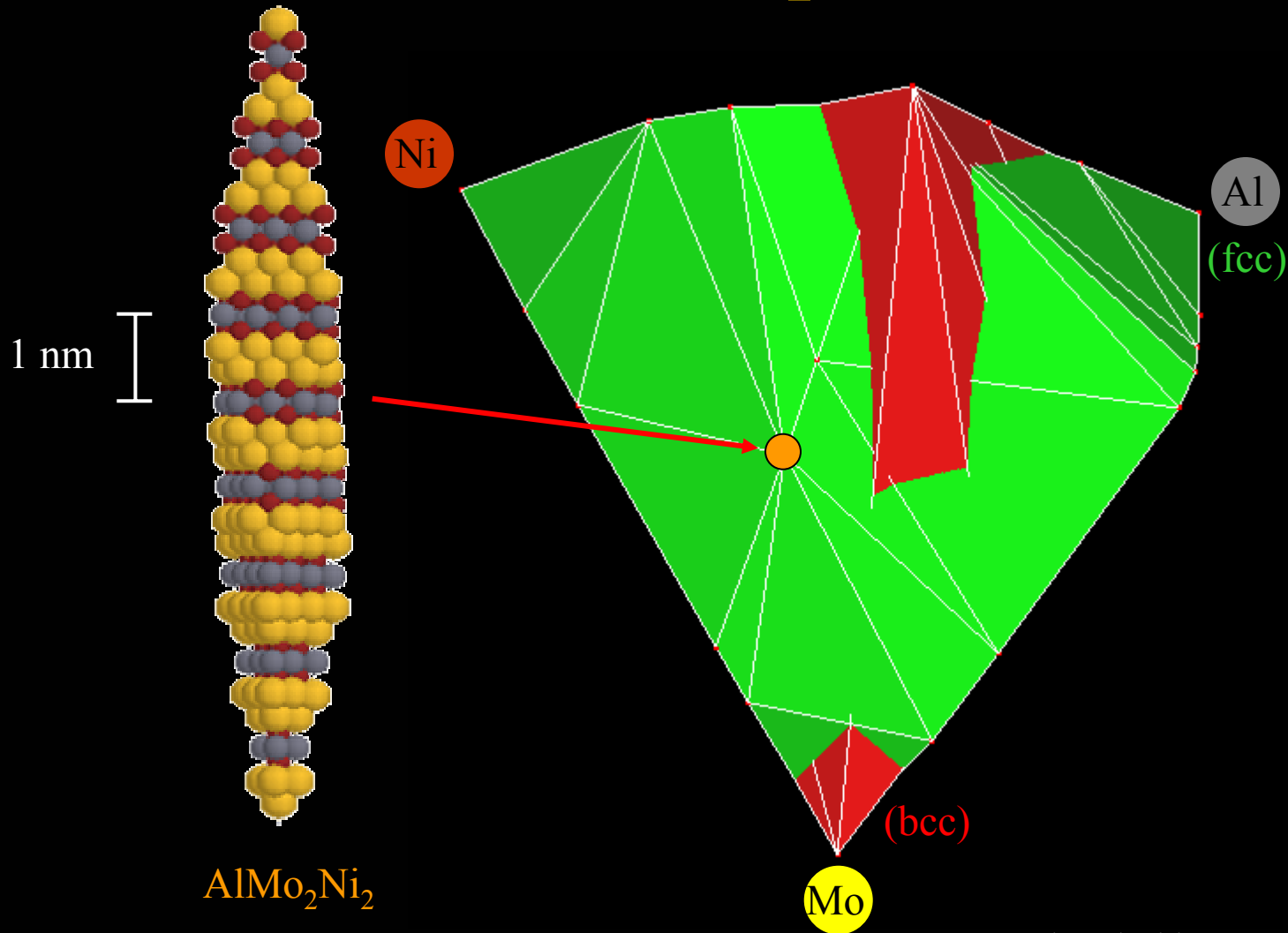
Hart, Blum, Walorski & Zunger, *Nat. Mat.* 4, 391 (2005)

# Search for new compounds in the Al-Mo-Ni system



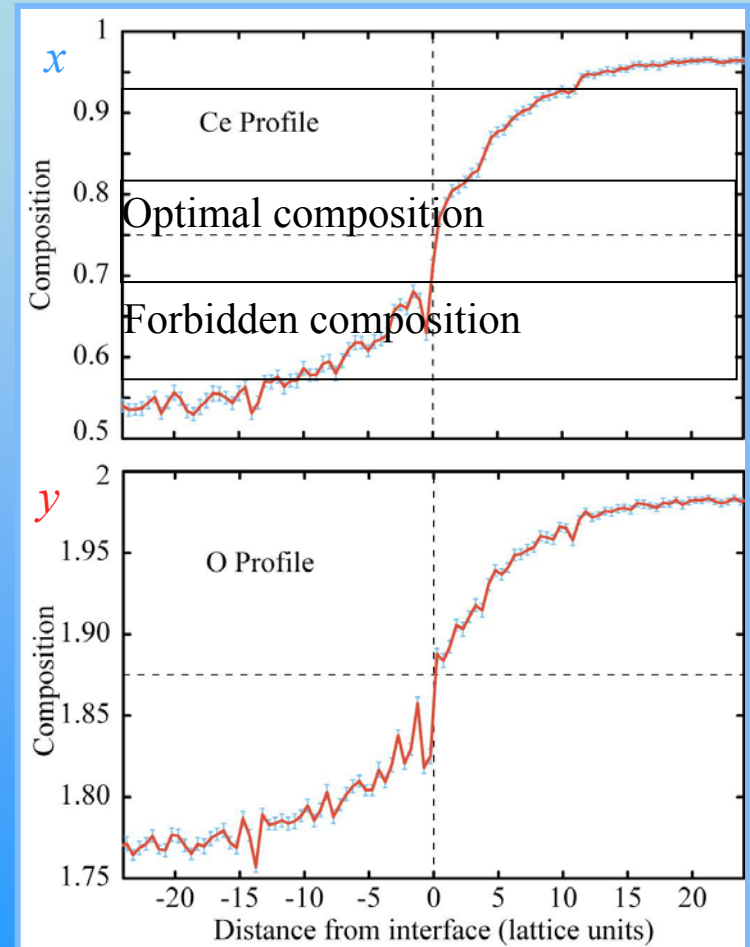
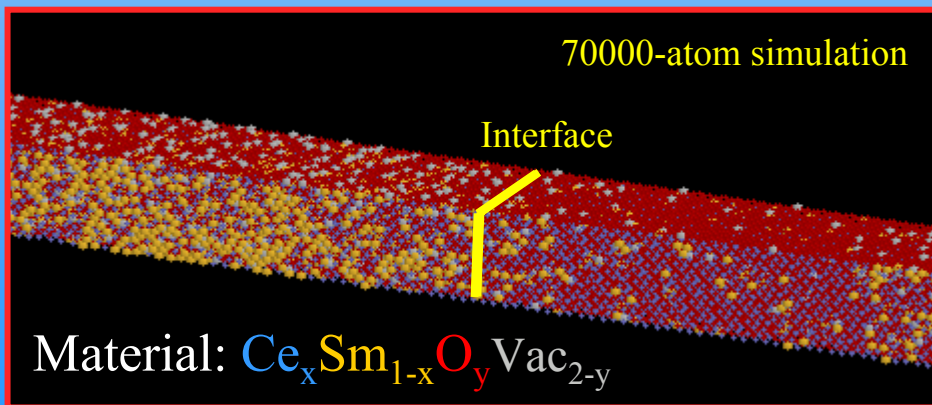
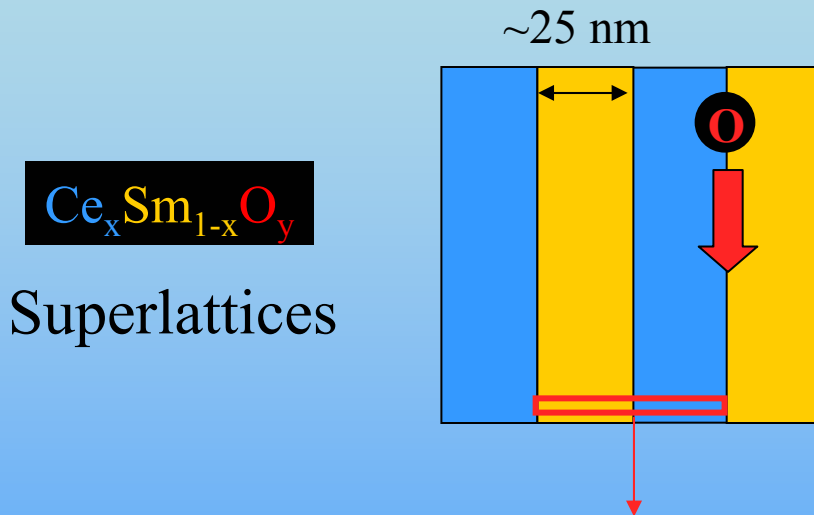


# Predicted Nano-Structured Compound

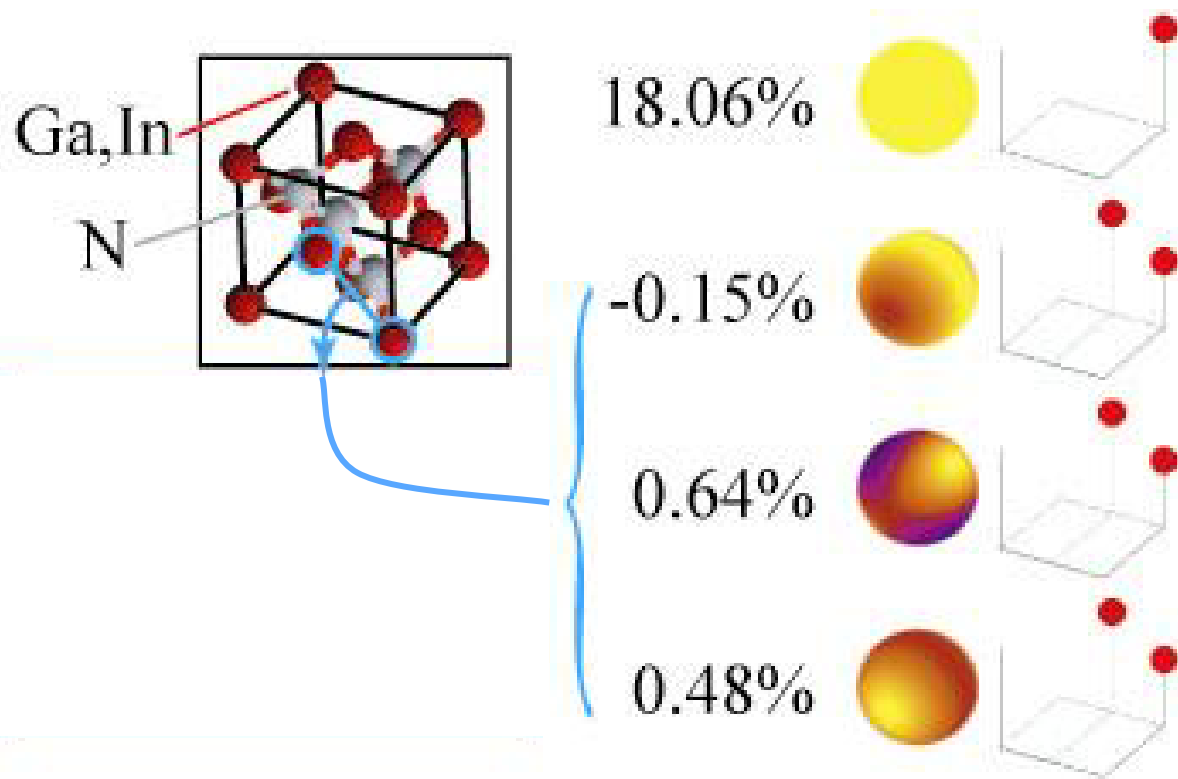


plotted with *MEDIT*, INRIA-Rocquencourt.

# Nanostructured Electrolyte (for fuel cell applications)

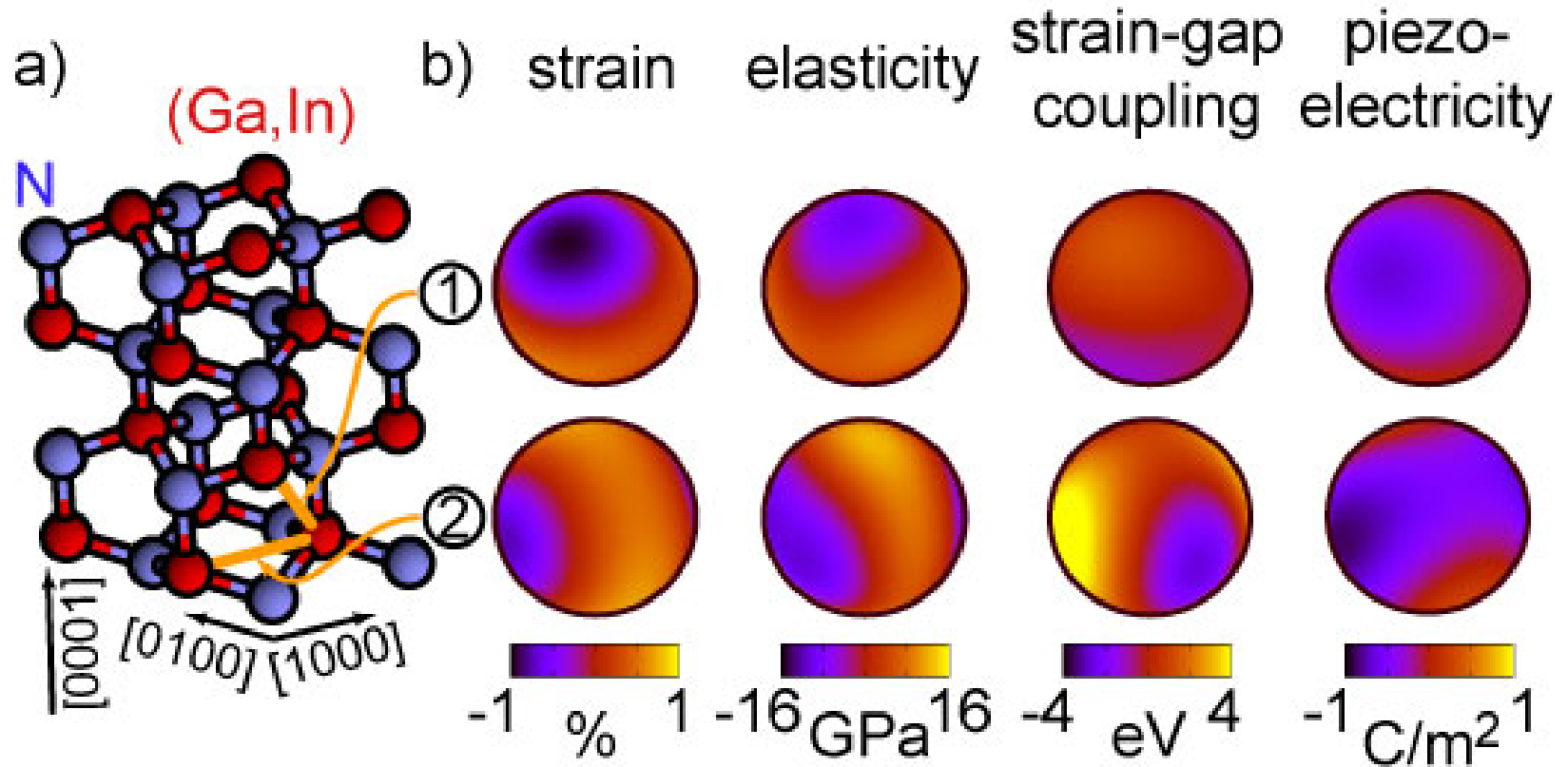


# First-principles determination of Structure-Property Relationships



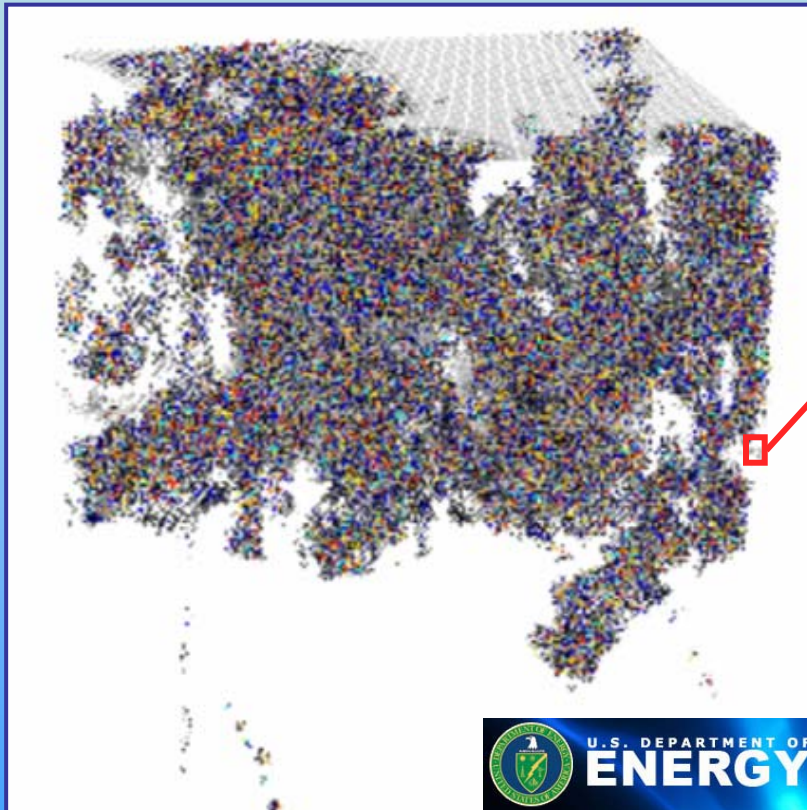
Example: Configuration-strain coupling

# More Structure-Property Relationship!

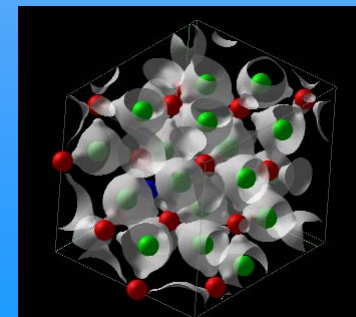
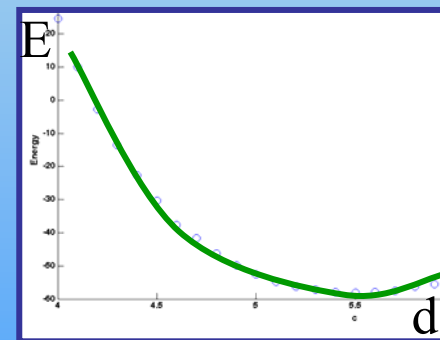
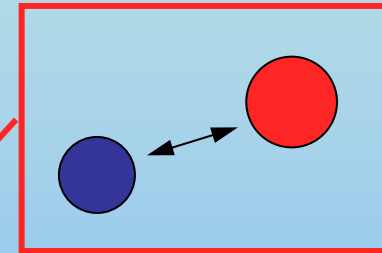


➔ Useful input for design and optimization of optoelectronic devices

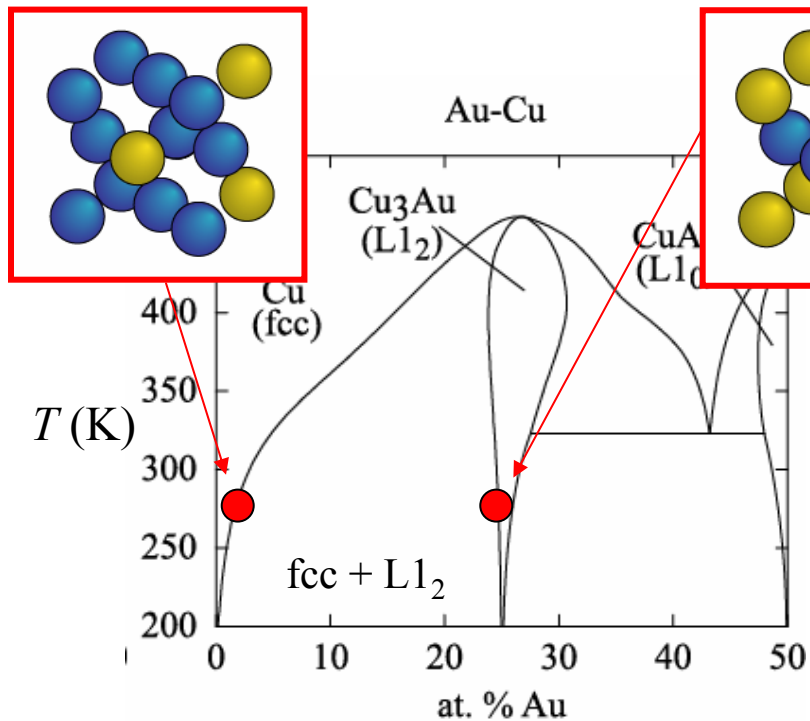
# Interatomic Potential Construction from Quantum-Mechanics



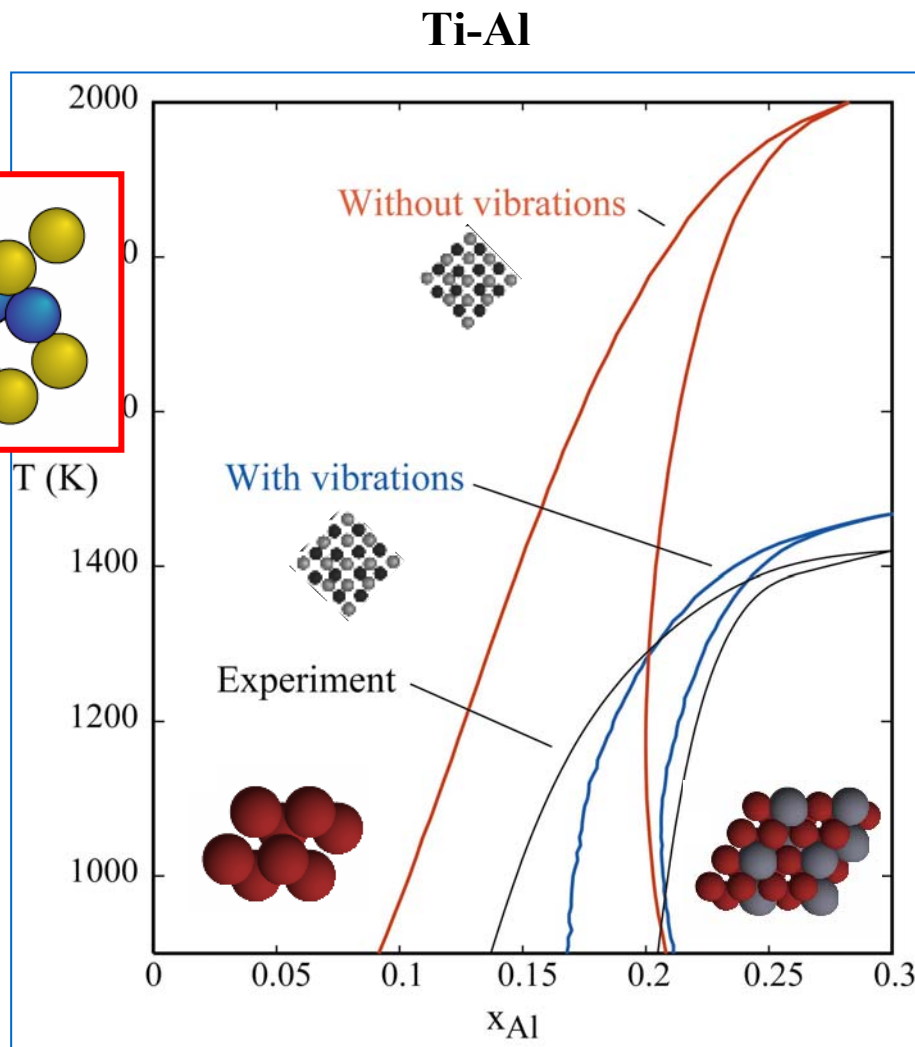
Radiation damage in nuclear fuel for advanced burner Reactors: modeling and experimental validation (PI: N. Jensen, UC Davis; Student: P. Tiwary)



# Calculated phase diagrams

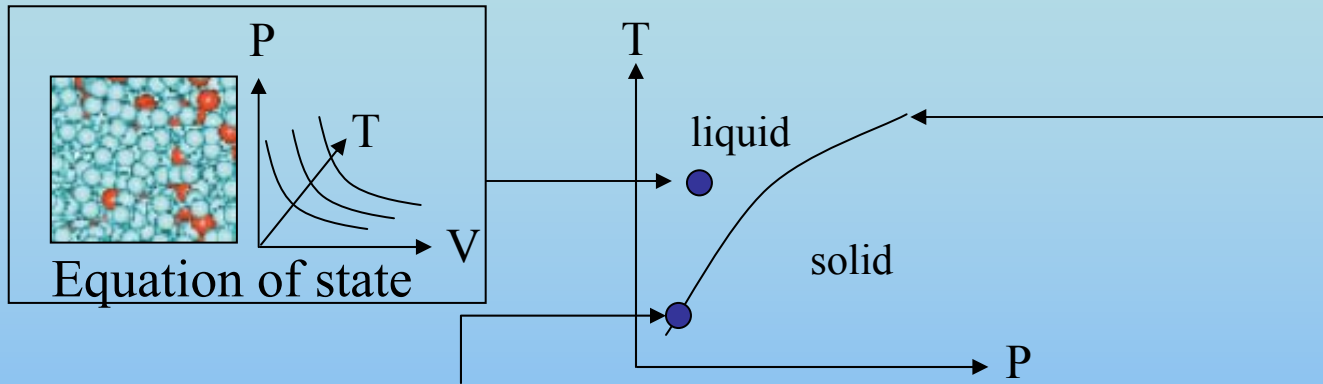


Alloy Automated  
**AM**  
Theoretic Toolkit



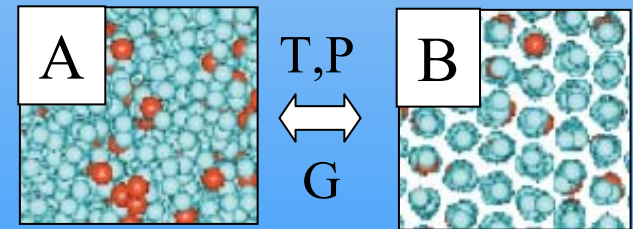
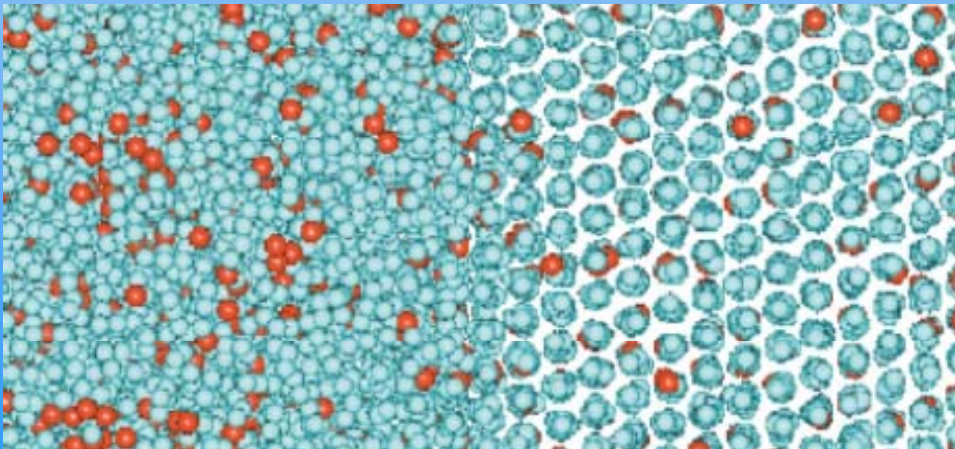


# Locating Phase Transitions and Equation of State (EOS) Calculations



Phase coexistence method

Free Energy Equalization



Boundary tracing:

$$\frac{dP}{dT} = \frac{1}{T} \frac{H^B - H^A}{V_B - V_A}$$

Ramalingam, Asta, van de Walle & Hoyt, *Interface Sci.*, 10: 149, 2002

PSAAP Center for the Predictive Modeling and Simulation of High Energy Density Dynamic Response of Materials (PI: M. Ortiz, Caltech).

